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FORM	A.O.	C.G.	E.O.	DEPT. OF MINES
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EXPLORATION LICENCE 23/79

WYNYARD, TASMANIA

REPORT FOR THE AREA RELINQUISHED 1ST DECEMBER, 1984

OPEN FILE

MICROFILMED

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- 2. Geology 1:100,000
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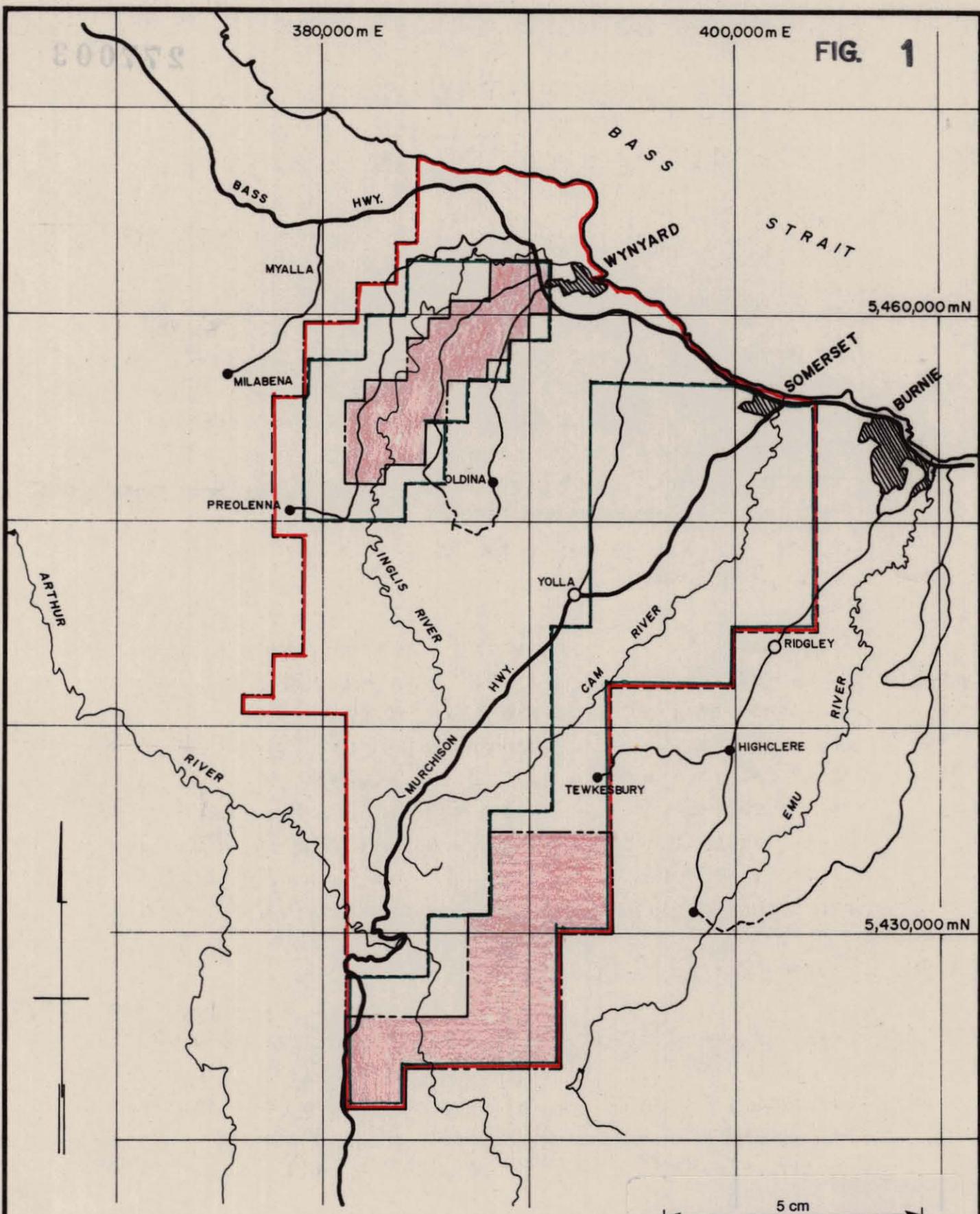
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APPENDICES

- 1. Stream Sediment and Pan Concentrate Sample Results
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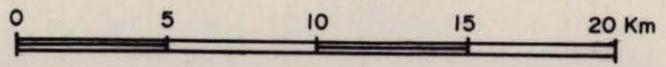
FIG. 1



- Original EL boundary
- From 1-12-83
- From 1-12-84

A.M.G. ZONE 55

SCALE 1: 250,000



Centre
Melbourne

Date
30-10-84

THE BROKEN HILL PROPRIETARY CO. LTD.
EL.23/79, WYNYARD, TASMANIA.

LOCATION MAP

Project No
T 64

Drawing No
A4-2473

EXPLORATION LICENCE 23/79WYNYARD, TASMANIAREPORT FOR THE AREA RELINQUISHED 1ST DECEMBER, 19841. GENERAL

Exploration Licence 23/79 of 715 square kilometres was granted to The Broken Hill Proprietary Company Limited on 14th December, 1979. The Licence area was reduced to 372 square kilometres on 1st December, 1983, made up to two separate blocks. In late August, 1984, notification was given of a further reduction to 124 square kilometres, this to be effective from 1st December, 1984. This report covers all exploration data relevant to the relinquished area.

Regionally the principal target was a massive sulphide hosted tin deposit of the Renison type. Work done in the relinquished portion of the licence area was restricted to aeromagnetic coverage and very limited geochemical sampling. The area is poorly prospective. In addition to the Tertiary basalt cover large areas also have a thick underlying sequence of Permian sediments, as well as local Jurassic dolerites. The prospective Cambrian and Precambrian lithologies are deeply buried.

2. GEOLOGY

Within the licence area extensive flows of Tertiary basalt and Permian sediments overlie a basement consisting largely of Precambrian and Cambrian rocks which form a major structural extension of sequences in the Mt. Bischoff, Cleveland area to the south west.

The oldest rocks present are the Precambrian Keith Metamorphics which occur in a belt 8-15km wide trending north east from Savage River to Wynyard. Rocks in this belt include pelitic schist, quartzite and minor

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amphibolite. Younger Precambrian rocks of the Burnie Quartzite and Slate Formation flank the Keith Metamorphics in the lower Cam River. Dolomite has not been recorded in the Precambrian rocks of the Wynyard area.

Cambrian sediments are exposed in a large window in Tertiary basalt, in the Hellyer River upstream from the Murchison Highway crossing. Lithologies present include red-brown lithicwacke, red shale and pyrite-bearing chert.

Ordovician sediments have not been located within the licence area. They flank a major anticline at Companion Hill to the east of the southern portion of E.L. 23/79.

Permian sediments, up to 300 metres thick, consist of a basal tillite unit, siltstone with thin oil shale and coal horizons, and sandstone.

Jurassic dolerite is exposed in the central part of the licence area where it invades both tillite and siltstone.

Tertiary rocks include basalt of highly variable thickness (locally believed to be up to 350 metres), as well as marine sediments, lacustrine clays, sands and gravels up to 60m in thickness.

3. WORK PROGRAMME

The following is a summary of work carried out:

1. Literature survey and review of data
2. Preliminary photogeological study and interpretation of Landsat image.
3. Reconnaissance geological mapping at 1:50,000 scale, with continuous updating as exploration proceeded.

4. Test Dighem II airborne EM survey.
5. High-resolution aeromagnetic survey covering the total licence area, with eastwest lines spaced at 250 metres and a mean sensor terrain clearance of 90 metres.
6. Stream sediment and pan-concentrate geochemical sampling over areas of basement exposure.
7. Orientation ground water sampling.

4. GEOPHYSICS

4.1 Aeromagnetics

Initially all available aeromagnetic data in the form of contour maps of total field intensity was compiled. In some instances trends due to major structures in basement rocks can be traced beneath the basalt and sedimentary rock cover.

A detailed aeromagnetic survey covering the entire licence area, with east-west lines spaced at 250 metres and a mean sensor terrain clearance of 90 metres, was flown in January, 1982. Coverage relevant to this report appears in Figure 3.

4.2 Dighem II EM

Four test lines of Dighem II were flown in the southern part of the licence area in early 1981. The location of the lines is shown on Figure 2.

Dighem II is a helicopter borne, frequency EM system which measures in In phase and Quadrature EM response in both a coaxial (standard) and coplanar (whaletail) pair of coils mounted in a 9 metre boom. The boom is towed at a nominal height of 36

metres at a speed of around 120 km/hr. A total field magnetometer is also carried with the system, the sensor being suspended at a height of around 50 metres.

Profiles of the measured data plus Dighem's computer generated interpretation functions were presented in the previous relinquishment report.

5. GEOCHEMISTRY

Figure 4 shows stream sediment, rock chip and water sample locations within the relinquished areas, while Figure 5 shows pan-concentrate sample locations with tin values.

Appendix 1 contains results of the stream sediment and pan-concentrate sampling programme. No anomalous values were obtained from the sieved samples, but a number of pan-concentrates confirmed the presence of tin within several drainage systems. Possible sources include Tertiary sediments, Permian tillite and Basement exposures.

It is considered probable that most of the tin is being derived from Tertiary sediments. If this is true large parts of the area sampled will thus be naturally contaminated and this greatly curtails effective anomaly recognition within the prospective Precambrian and Cambrian sediments.

Two quarries with Precambrian exposure were located at pan concentrate sample sites 100P and 118P. Both contained pyritic black shales and heavily veined sandstones and quartzites. Two rockchip samples gave relatively high tin values (55,190 ppm) but follow-up sampling failed to duplicate the earlier results.

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Several samples of pyrite rich veins and pods in quartzite float from the Cam River at site 124P were also analysed. These returned high arsenic values (450,700 ppm) but are probably pebbles washed out of the Permian tillites.

Analytical results and petrology relevant to the rock chip samples are given in Appendix 2.

Appendix 3 includes details of an orientation water-sampling programme carried out in 1982. Results were not encouraging and no follow-up was attempted.

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APPENDIX 1

STREAM SEDIMENT
AND PAN-CONCENTRATE SAMPLE

RESULTS

ANALABS

A division of MacDonald Hamilton & Co. Pty. Ltd.

Phone (09) 458 7999

52 Murray Road, Welshpool, W.A. 6106

Telex AA92560

ANALYTICAL REPORT No. 14.4 08 1169

THIS REPORT MUST BE READ IN CONJUNCTION WITH THE ACCOMPANYING ANALYTICAL DATA

E. H. P. Company
 GPO Box 11490
 Perth WA
 Telephone 08 3300

ORDER No.	PROJECT
605090	T 650
DATE RECEIVED	RESULTS REQUIRED
3.5.82	

NO. OF PAGES OF RESULTS	DATE REPORTED	No. OF COPIES	TOTAL No. OF SAMPLES
	5.5.82		17

STATE OF SAMPLES	SAMPLE NUMBERS	PRE-TREATMENT							ANALYSIS			
		DRY	CRUSH	SPLIT	PULV. VERSE	SEIVE	OTHER SEE REMARKS	NONE	REF. TO ANALYSIS SECTION	PREPARATION	METHOD	

1. Hs Above

REMARKS
 - 30#

STATE OF SAMPLES	ANALYSIS - PREPARATION	ANALYSIS - METHOD
whole core WC	perchloric acid A1	atomic absorption AAS
split core SC	hydrochloric acid A2	x-ray fluorescence XRF
cutting CU	nitric acid A3	spectrophotometry SPEC
rock Ro	aqua regia A4	colorimetry COL
soil SO	nitric-perchloric A5	chromatography CHR
pulp PU	HF mixture A6	titration IG
water WA	HF under pressure A7	other chemical means CHEM
tissue TI	fusion A8	miscellaneous MISC
stream sediment SS		fluorescence FLUOR
heavy mineral HM		inductively coupled plasma ICP

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ANALYTICAL DATA

SAMPLE PREFIX

REPORT NUMBER

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CLIENT ORDER No.

PAGE

		14.4 05 1150	5.5.82	005090	1 OF 2
TUBE No.	SAMPLE No.	Cu	Zn	Pb	
1	WYN 50	10	35	5	
2	WYN 51	5	40	5	
3	WYN 52	5	30	5	
4	WYN 53	X	25	5	
5	WYN 54	X	20	X	
6	WYN 55	10	30	5	
7	WYN 56	10	30	X	
8	WYN 57	5	35	5	
9	WYN 58	5	25	X	
10	WYN 59	10	40	10	
11	WYN 60	5	20	X	
12	WYN 61	10	45	5	
13	WYN 62	5	15	5	
14	WYN 63	10	50	5	
15	WYN 64	X	—	—	
16	WYN 65	10	15	30	
17	WYN 66	10	70	5	
18					
19					
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Results in ppm unless otherwise specified
 T = element present; but concentration too low to measure
 X = element concentration is below detection limit
 — = element not determined

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REPORT NUMBER

REPORT DATE

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		14.4 DE 1160		10.5.62		685070		3 OF	
TUBE No.	SAMPLE No.	Sn	P						
1	NYN 50	X	X						
2	NYN 51	X	X						
3	NYN 52	X	X						
4	NYN 53	X	X						
5	NYN 54	X	X						
6	NYN 55	X	X						
7	NYN 56	X	X						
8	NYN 57	X	X						
9	NYN 58	X	X						
10	NYN 59	X	X						
11	NYN 60	X	X						
12	NYN 61	X	X						
13	NYN 62	X	X						
14	NYN 63	16	X						
15									
16									
17									
18									
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22									
23									
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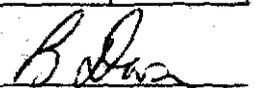
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TUBE No.	SAMPLE No.	Cu	Zn	Pb						
1	STD P54	215	850	105						
2	RPT N4N 50	10	80	10						
3										
4										
5										
6										
7										
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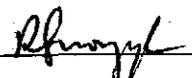
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1 OF 2

TUBE No.	SAMPLE No.	Cu	Pb	Zn					
1	T64/70	17	2	60					
2	T64/72	15	7	118					
3	T64/74	22	6	113					
4	T64/76	40	3	200					
5	T64/78	37	6	144					
6	T64/79	10	4	52					
7	T64/80	10	2	48					
8									
9									
10									
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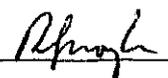
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PAGE

				14.4 08 1740 B	4.1.83	006627	3 OF 3	
TUBE No.	SAMPLE No.	Sn	W					
1	T64/70	4	X					
2	T64/72	X	X					
3	T64/74	15	X					
4	T64/76	6	X					
5	T64/78	7	X					
6	T64/79	X	X					
7	T64/80	7	X					
8								
9								
10								
11								
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23	DETECTION	3	10					
24	DIGESTION							
25	METHOD	402	401					

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 T = element present: but concentration too low to measure
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PAGE

SAMPLE PREFIX		REPORT NUMBER		REPORT DATE		CLIENT ORDER No.		PAGE	
		14.4 88 1741		06.01.83				1 OF 1	
TUBE No.	SAMPLE No.	Wt(g)	Sn	W					
1	T64-71P	74.22	6	X 4	/				
2	T64-73P	83.77	3	X 3	/				
3	T64-75P	49.23	75	X 37	/				
4	T64-77P	36.11	3	X 1	/				
5	T64-81P	22.96	1200	20 276	/				
6									
7									
8									
9									
10									
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21									
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23	DETECTION	.01	3	10					
24	DIGESTION								
25	METHOD		402	401					

Results in ppm unless otherwise specified
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SAMPLE PREFIX REPORT NUMBER REPORT DATE CLIENT ORDER No. PAGE

		14.4 08 1796			25.1.83		6610		1 OF 2	
TUBE No.	SAMPLE No.		Cu	Pb	Zn					
1	T64-83		56	24	161					
2	T64-85		31	32	220					
3	T64-87		33	20	102					
4	T64-89		8	19	29					
5	T64-90		9	18	40					
6	T64-91		18	22	76					
7	T64-93		20	14	98					
8	T64-95		45	19	65					
9	T64-97		23	52	69					
10	T64-99		34	22	110					
11	T64-101		14	26	41					
12	T64-103		20	17	90					
13	T64-105		11	21	34					
14	T64-107		25	12	77					
15	T64-109		28	25	93					
16										
17										
18										
19										
20										
21										
22										
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24										
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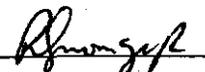
6610

1 OF 1

TUBE No.	SAMPLE No.	Sn	W						
1	T64-83	10	X						
2	T64-85	10	X						
3	T64-87	8	X						
4	T64-89	9	X						
5	T64-90	9	X						
6	T64-91	15	X						
7	T64-93	4	X						
8	T64-95	10	X						
9	T64-97	30	X						
10	T64-99	X	X						
11	T64-101	7	X						
12	T64-103	6	X						
13	T64-105	5	X						
14	T64-107	X	X						
15	T64-109	7	X						
16									
17									
18									
19									
20									
21									
22									
23	DETECTION	3	10						
24	DIGESTION								
25	METHOD	402	401						

Results in ppm unless otherwise specified
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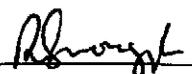
6609

1 OF 1

TUBE No.	SAMPLE No.	Sn	W	wt(g)						
1	T64-82P	20	X 4	18.07	/					
2	T64-84P	80	X 24	29.63	/					
3	T64-86P	95	X 19	20.05	/					
4	T64-88P	150	X 43	32.06	/					
5	T64-92P	8	X 4	50.20	/					
6	T64-94P	60	10 23	38.98	/					
7	T64-96P	190	X 59	31.21	/					
8	T64-98P	15	X 15	99.31	/					
9	T64-100P	10	X 3	30.85	/					
10	T64-102P	3	X 1	32.18	/					
11	T64-104P	40	X 13	31.61	/					
12	T64-106P	7	X 4	49.91	/					
13	T64-108P	360	10 77	21.45	/					
14	T64-110P	300	X 153	51.07	/					
15										
16										
17										
18										
19										
20										
21										
22										
23	DETECTION	3	10							
24	DIGESTION									
25	METHOD	402	401							

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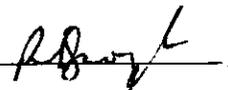
		14.4 08 1813				15.2.83		6613		1 OF 2	
TUBE No.	SAMPLE No.		Cu	Pb	Zn	Sn	W				
1	T64-111		40	12	128	6	X				
2	T64-113		19	10	113	8	X				
3	T64-115		47	5	155	X	X				
4	T64-117		30	9	125	4	X				
5	T64-119		25	8	113	5	X				
6	T64-121		35	5	123	8	X				
7	T64-122		51	7	123	4	X				
8	T64-123		24	4	122	6	X				
9	T64-125		33	8	154	3	X				
10	T64-127		32	14	137	5	X				
11											
12											
13											
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14.4 08 1812

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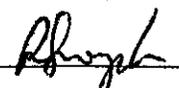
6612

1 OF 1

TUBE No.	SAMPLE No.	Wt(g)	Sn	W					
1	T64-112P	46.48	170	X 79	/				
2	T64-114P	60.83	150	X 91	/				
3	T64-116P	47.99	10	X 5	/				
4	T64-118P	47.93	15	X 7	/				
5	T64-120P	36.67	20	X 7	/				
6	T64-124P	43.24	7	X 3	/				
7	T64-126P	39.82	20	X 8	/				
8									
9									
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22									
23	DETECTION		3	10					
24	DIGESTION								
25	METHOD		402	401					

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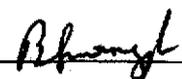
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PAGE

SAMPLE PREFIX		REPORT NUMBER		REPORT DATE		CLIENT ORDER No.		PAGE	
169		14.4 08 1877		24.3.88		006616		1 OF 1	
TUBE No.	SAMPLE No.	Wt (gr)	Sn	W					
1	129P /	60.89	45	X 27	/				
2	130P /	38.03	8	X 3	/				
3	131P	15.56	X	X					
4	132P /	59.80	5	X 3	/				
5	133P	79.00	X	X					
6	134P	70.02	X	X					
7	135P /	66.24	3	X 3	/				
8	136P /	57.71	3	X 2	/				
9									
10									
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22									
23	DETECTION		3	10					
24	DIGESTION								
25	METHOD		40%	40%					

Results in ppm unless otherwise specified
 T = element present; but concentration too low to measure
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CLIENT ORDER No.

PAGE

T64		14.4 00 1901			8.4.83		006620		1 OF 1	
TUBE No.	SAMPLE No.	Wt(gr)	Sn	W						
1	143P /	70.68	13	14 g	/					
2	144P /	50.84	387	X 194	/					
3	145P /	94.66	5	X 5	/					
4	146P	94.78	X	X						
5	147P /	66.17	3	X 2	/					
6	148P	61.78	X	X						
7	149P	105.29	X	22						
8	150P	67.76	X	X						
9	151P	95.63	X	X						
10	152P	103.05	X	X						
11	153P	72.19	X	X						
12	154P	89.74	X	X						
13	155P /	62.94	3	X 2	/					
14	156P	31.68	X	X						
15										
16										
17										
18										
19										
20										
21										
22										
23	DETECTION		3	10						
24	DIGESTION									
25	METHOD		400	401						

Results in ppm unless otherwise specified
 T = element present; but concentration too low to measure
 X = element concentration is below detection limit
 — = element not determined

 AUTHORISED
 OFFICER

APPENDIX 2

ROCK CHIPS - SAMPLE RESULTS,
PETROLOGY AND SAMPLE DESCRIPTION

SHEETS

SITE FLAGGED <input checked="" type="checkbox"/>		INTEREST NIL <input type="checkbox"/> SOME <input type="checkbox"/> DEFINITE <input type="checkbox"/>	
SAMPLE TYPE S.S. / P.C.		MAP/PHOTO NAME BURNIE 1:50,000	GRID REFERENCE 036504
		SAMPLE No. T64/Y2 T64/Y3P	
NATURE OF RIVER Flat, narrow. Rocky. Sheep Sth bank.		WATER Flowing. 8" deep.	VEGETATION Ti-Tree Blackberries ferns
PEBBLES		SITE	
Amount	Size	Lithology	Shape
		Basalt Qtzite	
		Rounding	Packing
			GOOD
		Silt %	HIGH
		Trap	Outcrop causing gravel traps in stream centre.
		Bottom	Gravel.
		Bedrock	QUARTZITE Basalt upstream.
		Assessment	Good.
FIELD COMMENTS		TENEMENT DETAILS	
Bedding 78°S Strike 100° Veins 90°, 45°		Coarse rounded heavy minerals, olive grains, fine grained sst/Qtzite, numerous milky qtz veins.	
		COOFE CREEK.	

Example of site description card filled out at each stream sample site.

REGION: NW TAS

PROJECT NO.: T640

PROSPECT: WYNYARD GENERAL GRID CO-ORDS:

LOCAL A.M.O.

DESCRIPTION

Sample Type: Rock

Rock Type:

Soil/Sediment Size Fraction:

WYN1: Lt. grey, pyrite rich chert (petrographic description available); float sample only
WYNA1-1: Lt green, emg. amphibolite, schistose texture, cut by qtz veins & minor limonite dissem. magnetite.

Project No: T640
Drawing No: A4-

ANALYSIS

Laboratory

ALS

Batch No

G219

Date Analysed

19/10/81

THE BROCKEN HILL PROPRIETARY CO. LTD.

GENERAL SAMPLE DESCRIPTION

Element	Sn	W	Mo	As	Au*	Ag	Cu	Pb	Zn	Ni	Co	Cr	Ba	Sr	Sb		
Method	XRF	XRF															
WYN1	<5	<10	<2	10	45	4	15	70	10	20	25	135	170	5	115		
WYNA1-1	<5	<10	2	3	50	1	15	20	15	30	15	30	N.R.	N.R.	<10		

REMARKS: * Au in ppb; N.R. - no result.

220

Logged or Sampled by: R. HINE

Date:

Cust: H-B ACT
Date: 26/11/81

277028

WYN1 (MRL 12870): Quartz-Pyrite Rock:

Composition: Fine to microgranular, incipiently sericite-stained quartz (60%), evenly disseminated fine sub- to euhedral pyrite. Sparse irregular quartz veinlets; sericite clots.

Fabric: Fractured/quartz-healed massive quartz-pyrite rock. Incipiently re-crystallized quartz. Pyrite mean 100 microns.

Accessories: Semi-pervasive, ultrafine, cloudy ?rutile.

Comments: Sinter-like facies (mildly recrystallized "pyritic chert").

Location: Float in stream between HEL 4 and HEL 5 sites.

CP1 (MRL 12871): Lithic Arkose:

Composition: Framework of angular to sub-angular quartz, subordinate oligoclase-albite, orthoclase, microcline; greenschist metapelite, chert-metaquartzite clasts. Sericitic, quartzo.

Fabric: Poorly sorted, essentially unbedded, silty fine to medium psammite.

Accessories: Conspicuous detrital biotite, garnet (almandine). Minor graphite, muscovite.

Comments: Polymictic sand with granitic and metasedimentary source, apparently weakly volcanomict, but ?felsite clasts sericitic, poorly resolved.

Location: Roadcut below basalt contact on South Basils Road.

CP2 (MRL 12872): Arkose:

Composition: Framework of splintery to angular quartz, alkali feldspar, subordinate metapelite, metagranite, ?metarhyolite clasts, biotite flakes. Sericitic, chloritic quartzofeldspathic.

Fabric: Similar to CP 1, but relatively well-bedded with silty partings, partly slumped.

Accessories: Detrital almandine, minor muscovite graphite flakes, rare cloudy rutile.

Comments: Similar and closely related to CP 1. Slightly finer-grained, relatively feldspathic. Similarly essentially unmetamorphosed. Weakly volcanomict.

Location: Exposure in small creek, 200m downstream from HEL 6.

029
CP3 (MRL 12873): "Vitric tuff" or "devitrified obsidian":

Composition: Frequent K-feldspathic spherulites in silicified, felsitic matrix with patchy microcrystalline albite, minor sericite.

Fabric: Spherulites mean 500 microns. Weakly flow-structured with faint relict ?eutaxitic microtextures.

Accessories: Sparse degraded/Fe-stained chlorite. Rare, very fine oxidised pyrite.

Comments: Conceivably a devitrified, thoroughly welded vitric tuff, but dependant on field relationships. Mildly altered (quartz-sericite-chlorite), unstressed.

Location: Float sample in Lockwood Creek downstream from CP2.

030

277031

ANALABS

A division of MacDonald Hamilton & Co. Pty. Ltd.

ANALYTICAL DATA

SAMPLE PREFIX

REPORT NUMBER

REPORT DATE

CLIENT ORDER No.

PAGE

14.4 08 1797

25.1.83

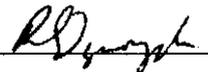
6611

1 OF 2

TUBE No.	SAMPLE No.	Cu	Pb	Zn	Hg				
1	T64-500	29	10	17	0.1				
2	T64-501	54	7	20	X				
3									
4									
5									
6									
7									
8									
9									
10									
11									
12									
13									
14									
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19									
20									
21									
22									
23									
24									
25									

Results in ppm unless otherwise specified
 T = element present; but concentration too low to measure
 X = element concentration is below detection limit
 — = element not determined

AUTHORISED OFFICER



031

277032

ANALABS

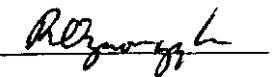
A division of MacDonald Hamilton & Co. Pty. Ltd.

ANALYTICAL DATA

SAMPLE PREFIX REPORT NUMBER REPORT DATE CLIENT ORDER No. PAGE

SAMPLE PREFIX		REPORT NUMBER			REPORT DATE	CLIENT ORDER No.			PAGE	
		14.4 08 1797B			28.1.83	6611			1 OF 1	
TUBE No.	SAMPLE No.	As	Sn	W						
1	T64-500	40	55	X						
2	T64-501	20	190	X						
3										
4										
5										
6										
7										
8										
9										
10										
11										
12										
13										
14										
15										
16										
17										
18										
19										
20										
21										
22										
23	DETECTION	2	3	10						
24	DIGESTION									
25	METHOD	401	402	401						

Results in ppm unless otherwise specified
 T = element present; but concentration too low to measure
 X = element concentration is below detection limit
 — = element not determined

AUTHORISED OFFICER 

032

277033

ANALABS

A division of MacDonald Hamilton & Co. Pty. Ltd.

ANALYTICAL DATA

SAMPLE PREFIX REPORT NUMBER REPORT DATE CLIENT ORDER No. PAGE

14.4.88 1814 15.2.88 6614 1 OF 2

TUBE No.	SAMPLE No.		Cu	Pb	Zn	Hg	As	Sn	W	
1	T64-502		20	140	145	X	X	3	X	
2	T64-503		25	105	125	X	X	3	X	
3	T64-504		25	15	55	X	X	X	X	
4	T64-505		25	25	45	X	50	4	X	
	T64-506		15	10	25	X	X	X	X	
6	T64-507		10	10	25	X	X	4	X	
7	T64-508		15	X	20	X	X	X	X	
8	T64-509		55	10	20	X	X	7	X	
9	T64-510		10	5	15	X	X	6	X	
10	T64-511		30	10	20	X	X	3	X	
11	T64-512		15	10	35	X	X	X	X	
12	T64-513		20	55	5	X	X	4	X	
13	T64-514		15	15	25	X	450	X	X	
14	T64-515		10	5	25	X	X	3	X	
15	T64-516		15	15	60	X	100	6	X	
16	T64-517		10	X	40	X	X	3	X	
17										
18										
19										
20										
21										
22										
23										
24										
25										

Results in ppm unless otherwise specified
 T = element present; but concentration too low to measure
 X = element concentration is below detection limit
 - = element not determined

AUTHORISED OFFICER *[Signature]*

REPORT CMS 83/2/14Samples MRL 14.423 - MRL 14.434

Twelve samples were received for thin-section preparation and petrological description; carbonate stain tests were carried out on the offcuts where necessary, and thin-sections were examined under both stereobinocular and petrological microscopes. Each sample is briefly described in the accompanying table.

Summary

Samples T64/900-902 are weakly metamorphosed clastic sediments; it is inferred that they contained significant amounts of feldspars, and hence have been classified as meta-arkoses. They are generally pyritic and carbonaceous, suggesting deposition under reducing conditions. T64/903 is an incipiently metamorphosed orthoquartzite thought to be related to the other three rocks. No Sn minerals were detected, but unless the levels were relatively high, detection would be pure chance; crushing of the rocks and concentration would be necessary to track down any Sn sources. Logically, the quartz veins could be potential sources, but they seem to be virtually monomineralic.

Samples WA 1-1 to 5 and 1-8 are igneous rocks ranging from olivine basalt to andesites and microdiorites; the andesites and microdiorites are certainly genetically related, but the olivine basalt is believed to be unrelated or only distantly so; it is a typical Tertiary basalt.

WA 1-6 and 7 are chemical sediments composed of pelletal chert and fine dolomite (with minor clastic components) in varying proportions.

H.W. Fander, M. Sc.

Sample No.	Rock Type - Composition	Fabric	Minor Minerals	Comments
T 64/900 (T.S. 45260)	<u>Meta-Arkose(?)</u> . A few larger quartz, chert grains in fine-sand-sized quartz, fibrous sericite after ?feldspar; leucoxic films throughout.	Good preferred orientation, almost schistose. Medium-grained.	Detrital zircon, tourmaline, leucocoxene. Some chert was dolomitic. Chalcedony veins.	Rock was originally a fine arkose or argillaceous siltstone/fine sandstone. Mildly metamorphosed only.
T 64/901	<u>Meta-Arkose(?)</u> . Stretched, stressed grains of quartz and chert, aggregates of fibrous sericite; interstitial pyrite, carbonaceous films, and sericite.	Strong preferred orientation, relict clastic textures. Medium-grained.	Many quartz veins, mostly barren; some carry fine sulphides.	Pyrite probably syngenetic. No Sn minerals recognised, but more detailed investigation required.
T 64/902	<u>Meta-Arkose(?)</u> . Stressed, partly recrystallized quartz grains, chert grains and sericite aggregates; many subparallel carbonaceous films.	Good preferred orientation. Medium-grained, well-sorted.	Small pyrite grains throughout. Carbon veins/layers. Quartz veins.	Fine ?pyrrhotite and ?chalcopyrite also occurs. Closely resembles 900, 901. Mildly metamorphosed.
T 64/903	<u>Stressed Orthoquartzite</u> . Framework grains dominantly stressed quartz, with minor chert; quartz cement, thin intergranular sericite films.	Well-sorted, weakly bedded, medium-sand size. All stressed.	Quartz veins, pre-dating stress. A few shale fragments. Carbon films, rare oxidised pyrite.	Probably related to the meta-arkoses. No indication of Sn minerals in rock (e.g. detrital cassiterite) or in veins.
WA 1-1	<u>Porphyritic Olivine Basalt</u> . Large phenocrysts of altered olivine, smaller labradorite and augite phenocrysts; groundmass of labradorite, pyroxene, dark glass, magnetite.	Subparallel alignment of phenocrysts and groundmass crystals.	Isolated small chlorite-filled amygdalae. Some olivine preserved.	Featureless, typical Tertiary basalt. Altered olivine is now bowlingite-iddingsite (soft, black).
WA 1-2	<u>Magnetite-Andesite</u> . Scattered andesine phenocrysts, groundmass of small andesine laths, interstitial magnetite and devitrified glass.	Random fabric, no flow-features. Crystals in three distinct size ranges.	Small chlorite pseudomorphs after ferromagnesian minerals. Quartz and carbonate veins.	White phenocrysts are andesine. Fabric suggests intrusive rock, rapidly cooled (quench textures).
WA 1-3	<u>Ferruginised Andesite</u> . Clusters of andesine phenocrysts in fine groundmass of andesine, magnetite, glass; bands of fine hematite impregnating rock.	Random fabric with a few phenocrysts. Quench textures in groundmass.	Scattered chloritised ferromagnesian minerals, including ?olivine.	Hematitic banding is post-magmatic and may be a type of "Liesegang Ring" effect. Rock correlates with WA 1-2.
WA 1-4	<u>Microdiorite</u> . Mostly stubby andesine crystals with interstitial magnetite, chloritised ferromagnesian minerals, secondary carbonate; a few andesine phenocrysts.	Uniform, random fabric. Medium-grained, weakly porphyritic.	Conspicuous carbonate-hematite-chlorite veins. Relict augite.	Probably a coarser-grained equivalent of WA 1-2/1-3. Minor intrusive; deuterically altered.

035

272036

REGION: PROJECT NO.: T640 PROSPECT: LYNBYRD GRID CO-ORDS: LOCAL A.M.O.

DESCRIPTION	Sample Type	Rock Type	Soil/Sediment Size Fraction		
T64/500	Rock Chip	Black shale, pyritic	from quarry	985496	Outcrop
SD1	"	Sandstone	black veins " stream	"	"
SD2	"	Black shale, pyritic	from stream	994436	"
SD3	"	"	" " Quarry	995436	"
SD4	"	Vein of haematite / silica	"	"	"
SD5	"	"	pyrite on fractures in sandstone from Quarry	"	"

ANALYSIS Laboratory ANALABS Batch No Date Analysed

Element Method Sample #	check															
	Cu	Pb	Zn	Ag	As	Sr	W		Sn							
SD0	29	10	17	0.1	40	55	-		80							
SD1	54	7	20	-	20	190	-		200							
SD2	30	140	145	-	-	3	-									
SD3	25	105	125	-	-	3	-									
SD4	25	15	55	-	-	-	-									
SD5	25	25	45	-	50	4	-									

REMARKS: 500 same rock as SD9. Re-analysed
 501 " " " SD6.

Logged or Sampled by: S. Kerber Date:

Project No:
Drawing No: AA-

THE BROKEN HILL PROPRIETARY CO. LTD.

GENERAL SAMPLE DESCRIPTION

Centre:
Date:

REGION: PROJECT NO.: T640 PROSPECT: WINYARD GRID CO-ORDS: LOCAL A.M.O.

DESCRIPTION	Sample Type	Rock Type	Soil/Sediment Size Fraction		
T64/ 506	Rock CHIP	Sandstone med. to coarse grained, black veining.	Report of 502	985496	O.C.
507	"	Phyllite: light tan, black veining.		"	O.C.
508	"	Shale pyritic, black with qz / calcite veins.		"	O.C.
509	"	Shale? siliceous, black, pyritic, veined.	Report of 508	"	O.C.
510	"	Sandstone siliceous, black isolated mineral.		"	O.C.
511	"	Shale: grey black pyritic, siliceous, calcite veins.		"	O.C.

ANALYSIS Laboratory ANALABS Batch No Date Analysed

Element	Cu	Pb	Zn	Ag	As	Sn	W	check	Sn								
506	15	10	25	-	-	-	-		4								
507	10	10	25	-	-	4	-										
508	15	-	20	-	-	-	-										
509	55	10	20	-	-	7	-		7								
510	10	5	15	-	-	6	-										
511	30	10	20	-	-	3	-										

REMARKS: 506 and 509 re-analysed.

Logged or Sampled by: S.P. KURBER Date:

Project No: Drawing No: A4-
 THE BROCKEN HILL PROPRIETARY CO. LTD.
 GENERAL SAMPLE DESCRIPTION
 Centre: Date:

037

277038

REGION:

PROJECT NO.: T640

PROSPECT: LYNBYARD

GRID CO-ORDS.:

LOCAL
A.M.O.

DESCRIPTION

Sample
TypeRock
TypeSoil/Sediment
Size Fraction

YOLLA 1.50,000.

T64/ 512 -	Rock Chip	Sandstone, grey siliceous, adjacent to pyrite band	946434	float
513 -	"	Pyrite vein, 8" wide line gneiss	"	"
514 -	"	Pyrite pods in sandstone pebble, rounded	945434	"
515 -	"	Sandstone siliceous, veined black mineral	946434	OC.
516 -	"	veined pyritic, qtz. pins boxworks	945434	float
517 -	"	"	"	"

ANALYSIS

Laboratory ANALABS.

Batch No

Date Analysed

Element Retained Sample #	Cu	Pb	Zn	Ag	As	Sn	W										
512	15	10	35	-	-	-	-										
513	20	55	5	-	-	4	-										
514	15	15	25	-	450	-	-										
515	10	5	25	-	-	3	-										
516	15	15	60	-	700	6	-										
517	10	-	40	-	-	3	-										

REMARKS:

Logged or
Sampled by: S.P. KERBER Date:

Project No:

Drawing No:

A4-

THE SHOCKER HILL PROPRIETARY CO. LTD.

GENERAL SAMPLE DESCRIPTION

Cont'd.

Date:

038

277039

APPENDIX 3

HYDROGEOCHEMISTRY

WARATAH AND WYNYARD E.L.'S

NORTHWEST TASMANIA

039

HYDROGEOCHEMISTRY

WARATAH AND WYNYARD EL'S

NORTH WEST TASMANIA.

G.WATMUFF

SENIOR GEOCHEMIST

SYDNEY.

WATER SAMPLING PROGRAMME AT WARATAH AND WYNYARD, TASMANIA.1. INTRODUCTION.

Much of Watatah and Wynyard Exploration Licences in Tasmania are covered by a relatively thin veneer of Tertiary basalt anywhere from 0 to 100 + m thick. The underlying topography appears gently undulating and covered by patches of clean Tertiary quartz gravel locally containing carbonized wood fragments. This appears no more than a few meters thick in the Waratah area, but may be several tens of meters thick locally in the Wynyard EL south of Table Cape. Windows of older Cambrian and Pre cambrian rocks appear through the basalt forming low silicified hills in the Waratah area, rising above the basalt plain.

The target, tin mineralization of the Mt. Bischoff or Cleveland or Renison type, is sought in the Cambrian and Pre cambrian rocks below the gravel/basalt cover. Present geophysical techniques are severely hampered in their effectiveness to detect mineralization under these conditions. Magnetic bodies will be masked by the strongly magnetic basalt. Aeromagnetic data may best serve to indicate the thin areas of basalt which possibly overlie topographic highs. These may represent more resistant silicified and ? mineralized rocks. Electrical methods may be effected by ground water which is encountered at or within a few meters of the surface the year round.

It was considered therefore that ground water sampling may be a useful approach to get information about the rocks below the basalt. A proposal was put to sample spring water around the scarp of the basalt plateau below the basalt/Cambrian-Pre cambrian sediments contact.

...../2

2. METHODS.

2.1 Time of sampling.

Sampling was carried out in March and April of 1982 at the end of the dry season (monthly rainfall distribution for three NW region centres is given in figure 1). This time was chosen because :

- 1) dilution of spring water from rainfall run-off was likely to be at a minimum,
- 2) the recharge rate is at its lowest thereby producing the highest salt content for the spring water and,
- 3) The springs still running at the end of the dry period are more likely to represent longer and therefore more significant groundwater flow paths than those springs that might run for a couple of days after a rain period.

2.2 Sampling Methods.

The samples were collected in one litre low density polyethylene bottles with plastic screw caps. Three samples were taken at each site, one only being acidified. The acidified and one non acidified samples were despatched to Allison Laboratories Ltd in Hobart and the other non acidified sample was retained for analysis at field base.

The bottles were rinsed with the sample twice in most cases before filling and where possible the bottles were completely filled. When filling, the mouth of the bottle was pointed in the direction from which the water flowed and filled slowly to avoid backwash contamination from the sampler's hand (a source of Na, Cl, Zn and possibly As). Where depth and rate of flow was very low a hole was dug by hand deep enough to immerse the bottle for filling. When the water had completely cleared of visible clay suspension the pool was deemed flushed and the sample carefully taken using a plastic glove to cover the sampler's hand.

2.3 Bottle Preparation.

The bottles for sample collection were prepared by East Melbourne Laboratories Pty Ltd in Melbourne and transported by sea and road to field base camp. The acidified sample bottles were acid washed (method 3.2 E.P.A. (1979)) and a glass vial (also acid washed) containing 10ml of double distilled concentrated HNO₃ was placed in each bottle. The non-acidified sample bottles were given a normal wash (method 3.1 EPA (1979)).

The use of a glass vial of pure HNO₃ placed in the bottle prior to going into the field has been shown to reduce the risk of contamination considerably. (Guest and Blutstein (1981)). It is obviously more convenient (and much less dangerous!) than carrying a separate dispenser of concentrated HNO₃ on a difficult bush traverse. The glass vials are simply broken inside the bottles immediately after they have been filled with the sample and sealed by giving the portion of the bottle in contact with the vial a sharp tap from the outside. The bottle is then shaken to disperse the acid.

2.4 Site Analyses.

Four parameters were determined on site: water temperature, conductivity, pH and dissolved oxygen. Conductivity was determined using a TPS LC 81 conductivity meter and K = 1 electrode. pH and dissolved oxygen were determined with an Orion 407 A specific ion meter using an Orion model 91 - 62 glass electrode (pH) and an Orion model 97 - 08 oxygen electrode. The electrodes were calibrated on site before each measurement.

The oxygen electrode is temperature compensating giving corrected dissolved oxygen values even though the temperature of the standardizing medium (water saturated air) is different from the sample temperature. The pH electrode is not temperature compensating but the difference between

standards and unknowns did not exceed 10°C and was usually about 5°C. This will produce an error of no more than about 0.05 pH units in the temperature range encountered.

The conductivity electrode was calibrated on 25.4 ppm and 254 ppm NaCl solutions (reading 53.8 and 520 μS at 25°C respectively) before going into the field. Only the 25.4 ppm NaCl calibration was used as most waters were found to have conductivities between 30 and 170 μS. All conductivity values are temperature corrected to 25°C. Other notes made at each site included an estimation of the average width and depth of the water flow, vegetation, types of rock fragments in the stream bed and any visible outcrop.

2.5 Field Base Analyses.

Five parameters were determined at base camp - pH, HCO₃⁻, Ca²⁺, Na⁺ and Cl⁻. In addition the temperature for the HCO₃⁻ determination was noted. The pH was too low (< 8.3) in all cases for any significant CO₃²⁻ species to be present.

Bicarbonate was determined by plotting potentiometric titration curves for each sample. The Orion 407 A specific ion meter and model 91 - 62 glass pH electrode was used and the titration was carried out on 100ml of unknown solution using a 10ml microburette delivering 0.0207 N H₂SO₄. In each case the titration was carried to below pH 4.5. The inflection point usually occurring at about pH 5.0 was noted and the amount of acid consumed used to calculate the HCO₃⁻ content.

Calcium, sodium and chloride were all determined using appropriate specific ion electrodes. Complexing was considered likely in the case of Ca²⁺ and therefore the standard addition method was used. Calcium

will form complexes with HCO_3^- and SO_4^{2-} which cannot be detected with the Ca electrode using the direct reading method. Only free Ca^{2+} ions are detected. It was found in the two cases checked that the standard addition method gave values about 10% higher than the direct method. This proportion of complexing is typical for solutions of this concentration.

Sodium and chloride do not form significant complexes under normal solution conditions. Potentiometric titration for dissolved CO_2 using 0.0454 N Na_2CO_3 was carried out on a few samples and as expected yielded levels closely approximating those of water in equilibrium with the atmosphere. None of the samples could be collected close enough to the point of groundwater emergence to give values of the true ground water levels.

3.0 RESULTS.

3.1 Site and Field Base Analyses.

Site and field base analyses are summarized on the analysis sheets (Appendix 1).

3.1.1 Water Temperature.

Field water temperatures are significantly below maximum daily air temperature by between 5 and 10°C and most commonly in the range of 10 to 12°C.

3.1.2 pH.

Field pH is mostly in the range 6.3 to 7.6. Four samples occurred in the range 4.7 to 6.0, but each had oxygen levels below those for water in atmospheric equilibrium and flow cross-sections less than

5cm² (figure 2). It is likely that this water is in various degrees of stagnation and has significant humic acid content. Even within the common range of pH the deviation from saturation with atmospheric oxygen is broadly related to pH. The median pH for water at atmospheric equilibrium is about 7.2. Water containing 0.75 ppm oxygen less than saturation has a median pH of 6.6 (figure 3).

The difference between base camp determined pH and field site determined pH is 0.05 ± 0.19 (1σ) pH units, the field site determinations being the lower on average.

3.1.3 Conductivity.

Conductivities ranged between 30 and 170 μ S with the exception of 007W, mine adit water from Mt. Bischoff. Here the conductivity was extremely variable, depending where the electrode was placed in the pool of water.

3.1.4 Free CO₂.

Free CO₂ was determined on 8 samples and these generally agreed with the nomographically determined values at base camp temperature, pH and HCO₃⁻. The determinations were carried out potentiometrically using 0.0454 M Na₂CO₃. The determinations were discontinued as it was considered that the CO₂ content of the water would be at equilibrium with the atmosphere and any deviation was likely to be biological rather than a reflection of original groundwater levels.

3.1.5 Calcium, Sodium, Chloride.

Calcium ranged from 1ppm to 7.2ppm and there is no detectable difference between the Waratah and Wynyard samples. In contrast Na is higher in the Wynyard streams (11.5 to 23 ppm) compared to Waratah (4.5 to 9.9 ppm)

possibly reflecting the influence of sea spray.

Although chloride determinations were not completed for the Wynyard samples a similar area bias is suggested by the available chloride and conductivity data.

There is no apparent correlation between rock type and Na/Ca or Ca/Conductivity ratios or absolute Ca or Na values in either exploration licence. Hence it is difficult to relate the water back to rock type.

3.2 Laboratory Analyses.

The samples were all submitted to Allison Laboratories Pty Ltd in Hobart for analysis. The non acidified samples were analysed for Na, K, Cl', SO₄"', HCO₃'', F', pH, conductivity and in most cases Ca and Mg. The acidified samples were analysed for Cu, Pb, Zn, Fe, Ca, Mg, F, As and in some cases Na. The analyses were carried out generally in accord with APHA "Standard Methods for the Examination of Water and Waste Water" 14th Edition (1975).

Corresponding laboratory and field analyses are shown on the sample analysis sheets (Apper.dix 1).

3.2.1 pH

The pH values show no more than 0.5 pH units difference between field and laboratory determinations and are mostly within 0.1 or 0.2 pH units of each other.

3.2.2 Conductivity.

The field conductivity measurements are about 10% higher than the

laboratory conductivity measurements over the entire range (30 to 170 microsiemens (uS) @ 25°C) (fig.4). This suggests a calibration difference. The field instrument was calibrated at 53.8 uS for a 25.4 ppm NaCl solution and an automatic temperature compensation coefficient of $-2.2\%/^{\circ}\text{C}$.

3.2.3 Sodium.

Sodium shows an absolute error of 1.6 ppm over the range 3 to 21 ppm (fig.4). This may be explained by the presence of 1.6 ppm Na in the distilled water used to prepare the standard solutions for the field determinations.

3.2.4 Calcium.

Agreement for calcium is poor (fig.4) variation being anywhere between a factor of 1 and 2. Agreement between Ca determined on acidified and non-acidified sample pairs determined by the laboratory is much better, but leaves a little to be desired - of the 14 determinations, 11 non acidified samples returned higher values than their respective acidified samples and 2 were lower. This problem needs further investigation.

3.2.5 Chloride.

Field chloride determinations are higher than the laboratory by between 10 and 30% in most cases (fig.4). This is somewhat puzzling as the chloride electrode showed good response time and little tendency to drift. Chloride does not have complexing tendencies and is not significantly affected by even several months of storage so the variation is not readily explicable.

3.2.6 Bicarbonate.

The bicarbonate values show fairly good agreement. A laboratory vs field plot (fig.4) shows a linear array of points which is slightly deviant from

the 1:1 line.

3.2.7 Potassium and Magnesium.

Potassium and magnesium range from 0.3 to 1.3 ppm and 0.9 to 5.7 ppm respectively. No correlation is apparent between rock type and K or Mg content. Values for both species are clearly higher in the Wynyard EL than Waratah EL.

3.2.7 Base Metals : Copper, Lead, Zinc.

Values are generally below detection for Cu (1 ppb) and Pb (10 ppb). Zinc gave a background range of 2-7 ppb. The acid blank contained below-detection Cu and Pb and 2 ppb Zn.

Apart from the mine adit sample (007W) from Mt. Bischoff, the only sample anomalous in any of the base metals was 029W from the Wynyard EL, with 84 ppb Zn. This value is an order of magnitude higher than the upper limit of the background range and should be followed up with more detailed sampling.

Lead is below saturation with respect to $PbCO_3$ and $PbSO_4$ in all samples and similarly Cu and Zn are below saturation with respect to their sulphates, simple carbonates and malachite.

3.2.8 Iron.

Values range from 0.08 to 1.3 ppm with the exception of the mine adit sample at 14 ppm. The iron is in excess of that expected if it was in equilibrium with ferric hydroxide. It therefore appears that most of the iron detected is in the ferrous state and would be well below saturation with respect to ferrous hydroxide.

There is no relationship between stream flow section or pH and iron content, nor is there any relationship between rock type and Fe/conductivity ratio.

3.2.9 Fluoride.

Levels ranged from 0.05 ppm to 0.25 ppm in the natural stream water. A value of 1.2 ppm was obtained for the Mt. Bischoff mine adit sample. All values were below CaF_2 saturation by one to three orders of magnitude. No significant difference was noted between the values from the two EL's and no relationship between rock type and F' level was observed.

3.2.10 Arsenic.

Arsenic has been determined on six samples and because the mine adit sample (0.14 ppm) was in excess of the detection limit (0.02 ppm), no further analyses have been carried out until a more sensitive method is devised.

3.2.11 Sulphate.

All values are below detection (0.2 ppm).

Apparently genuine analytical difficulties exist for lowering the detection limits of the latter two species. Normally As can be determined down to 1-2 ppb but for some reason poisoning of an analytical reagent (Zn) seems to be occurring. Barium chloranalate is used to provide a very sensitive technique for SO_4^{2-} determination, however this chemical has not been obtainable in Australia and shipment from overseas is awaited. The classic turbidometric method is limited to about 100-200 ppb SO_4^{2-} detection.

3.3 Water Characterization.

The waters could not be charaterized in relation to rock type. Plots of K, Na-Fe-Mg,Ca ; Na-K-Ca; Fe-K-Ca and Ca-Fe-Mg (figs.5,6,7) revealed no meaningful separation of waters sampled from streams containing dominantly sediment and dominantly basalt fragments.

3.4 Contamination.

A contamination study was made for the trace components. An acid blank was analysed and returned $< 0.001 \text{ } \mu\text{Cu}$, $< 0.01 \text{ } \mu\text{Pb}$, $0.002 \text{ } \mu\text{Zn}$, $0.004 \text{ } \mu\text{Fe}$.

Ca, Mg and Na were not detected. Another acidified sample was sent which had been deliberately contaminated by pouring distilled water over human hands and allowing the water to run directly into the sample bottle.

This sample returned $0.001 \text{ } \mu\text{Cu}$, $< 0.01 \text{ } \mu\text{Pb}$, $0.005 \text{ } \mu\text{Zn}$ and $0.008 \text{ } \mu\text{Fe}$.

Mg and Ca were less than 0.1 ppm (detection).

3.5 Charge Balance.

A complete charge balance for all major species is given in Table 1.

The figures are calculated from the values obtained by Allison Laboratories.

No significant SO_4^{2-} species have been detected by the turbidimetric method, however the high net positive charge balance suggests the presence of some negative species. An improved analytical technique may show this indeed to be sulphate although the presence of nitrate cannot be ruled out.

4.0 Conclusions.

A real Zn anomaly of 84 ppb exists at sample site 029W in the Wynyard EL and should be followed up. No other anomalous results were obtained and all species with the exception of Fe appear to be at sub saturation levels.

REFERENCES.

1. Environmental Protection Authority (E.P.A.) of Victoria.
1979 : A Guide to the Sampling and Analysis of Water and Wastewater.
Publication 95/79. 42p.

2. Guest, R.L. and Blutstein, H.
1981 : Delivery of Acid Preservative for Trace Metal Determinations
in Waters. Analytical Chemistry 53, 727 - 731.

TABLE 1 :

FILE DESCRIPTION : LABORATORY DETERMINATIONS SURFACE WATER WARATAH/WYNYARD EL's TASMANIA

Sample no. Fe2+(epm) Ca2+(epm) Mg2+(epm) K+(epm) Na+(epm) Cl-(epm) HCO3-(epm) F-(epm) SO4--(epm) NETT

26	0.01	0.07	0.30	0.02	0.68	0.62	0.15	0.01	0.00	0.30
27	0.02	0.03	0.28	0.02	0.71	0.59	0.05	0.01	0.00	0.41
28	0.05	0.07	0.35	0.03	0.64	0.37	0.49	0.01	0.00	0.27
29	0.01	0.09	0.42	0.03	0.82	0.59	0.33	0.01	0.00	0.44
30	0.02	0.09	0.35	0.03	0.90	0.62	0.40	0.01	0.00	0.35
31	0.01	0.19	0.37	0.03	0.52	0.37	0.45	0.01	0.00	0.30
32	0.02	0.26	0.47	0.02	0.73	0.65	0.39	0.01	0.00	0.45
33	0.02	0.06	0.25	0.01	0.64	0.65	0.05*	0.01	0.00	0.37
34	0.02	0.11	0.31	0.02	0.55	0.45	0.28	0.01	0.00	0.28
35	0.01	0.20	0.40	0.03	0.46	0.34	0.57	0.01	0.00	0.18
36	0.01	0.17	0.40	0.03	0.74	0.54	0.51	0.01	0.00	0.30

* : H⁺

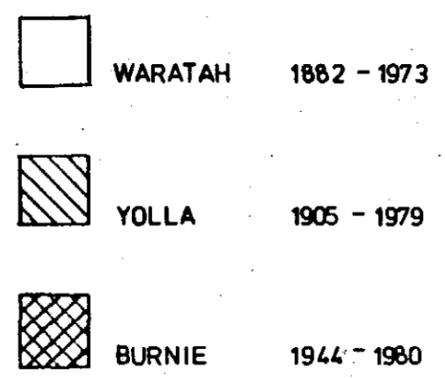
+ : Zn²⁺ + Cu²⁺ included

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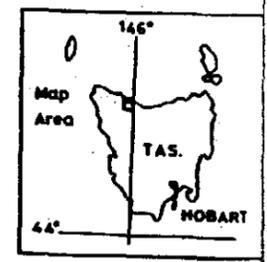
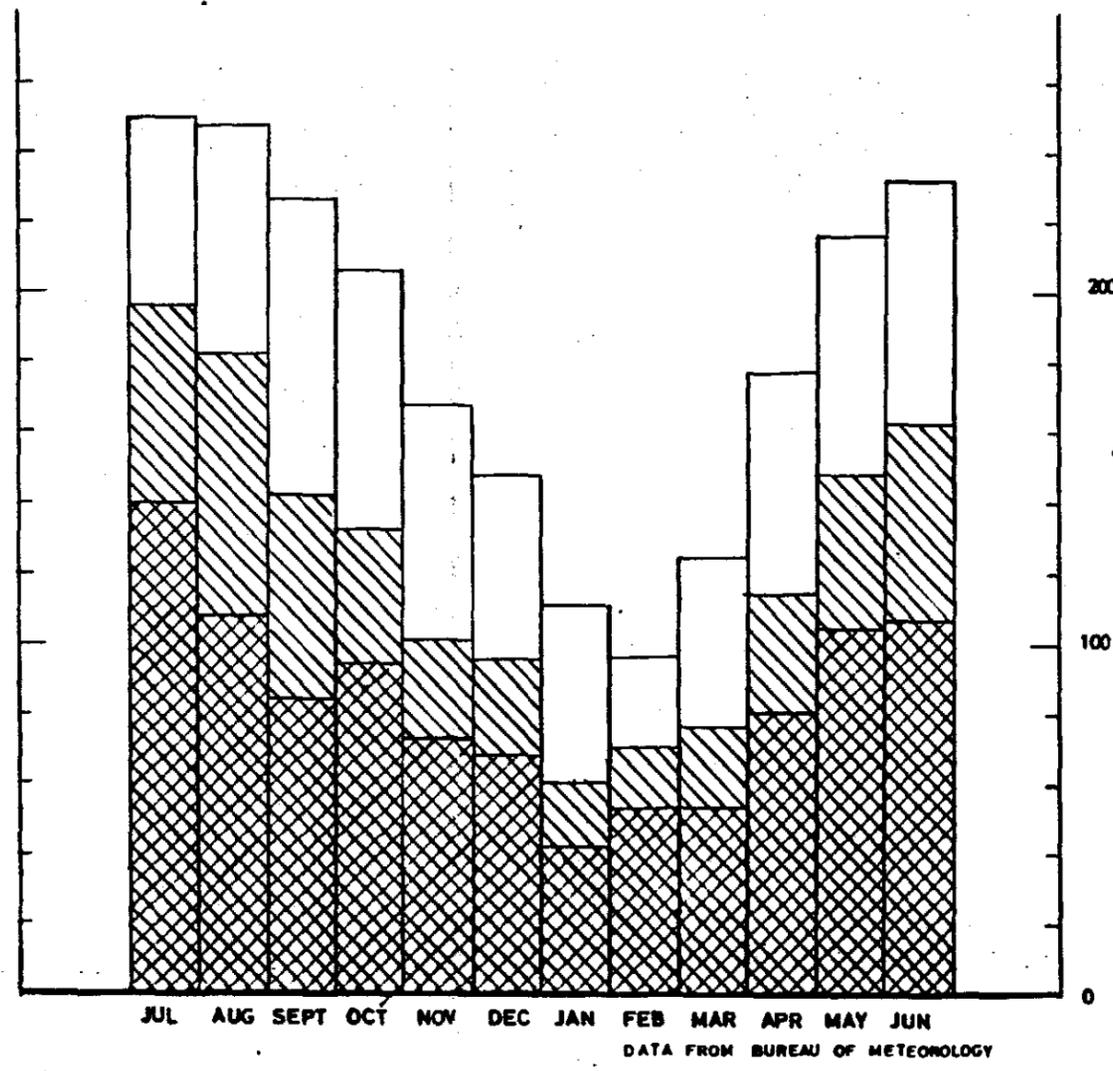
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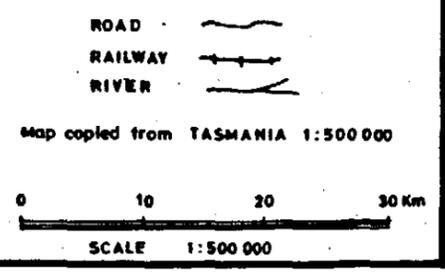
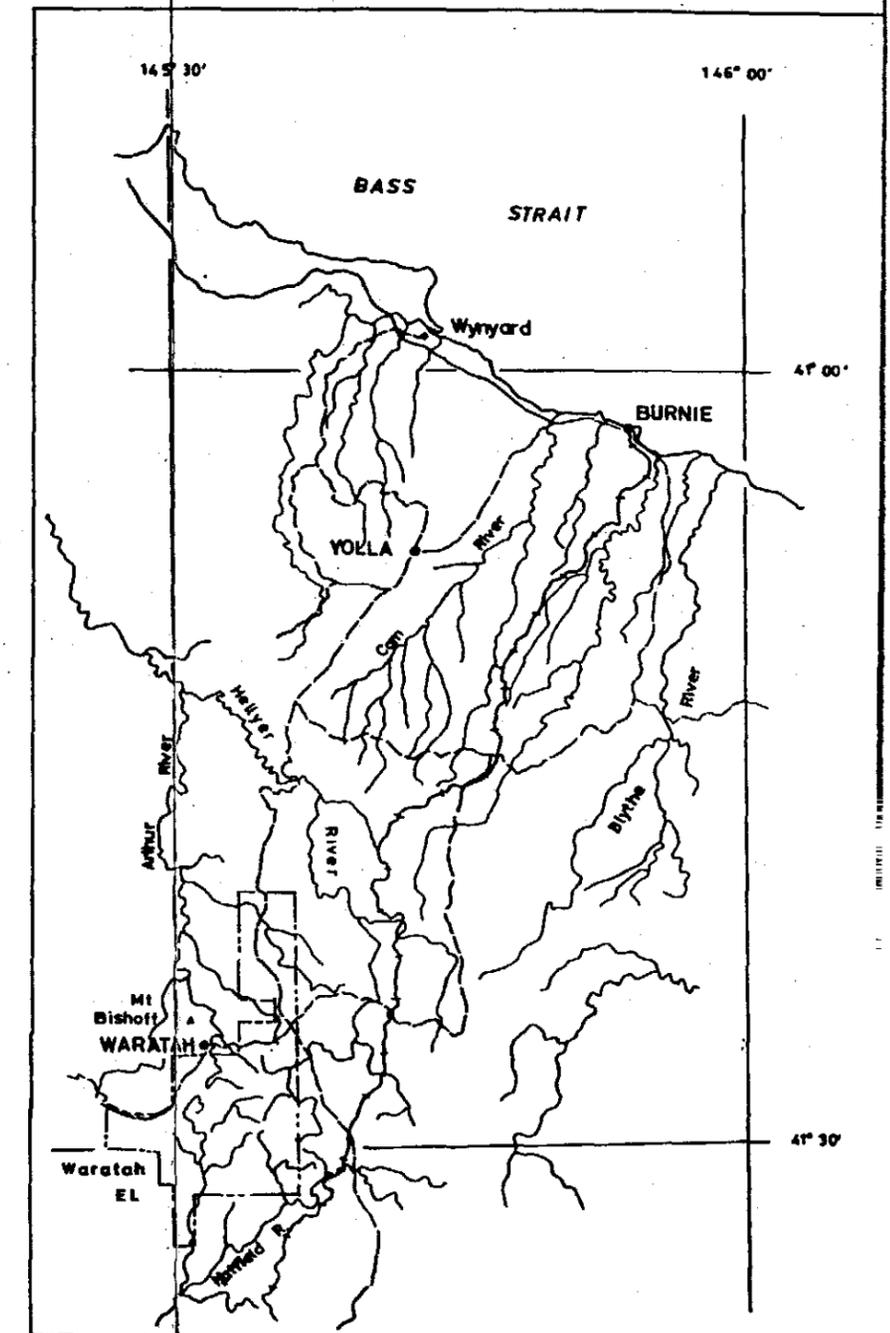
FIG. 1



Graph of Average Monthly Rainfall
WARATAH - YOLLA - BURNIE DISTRICT

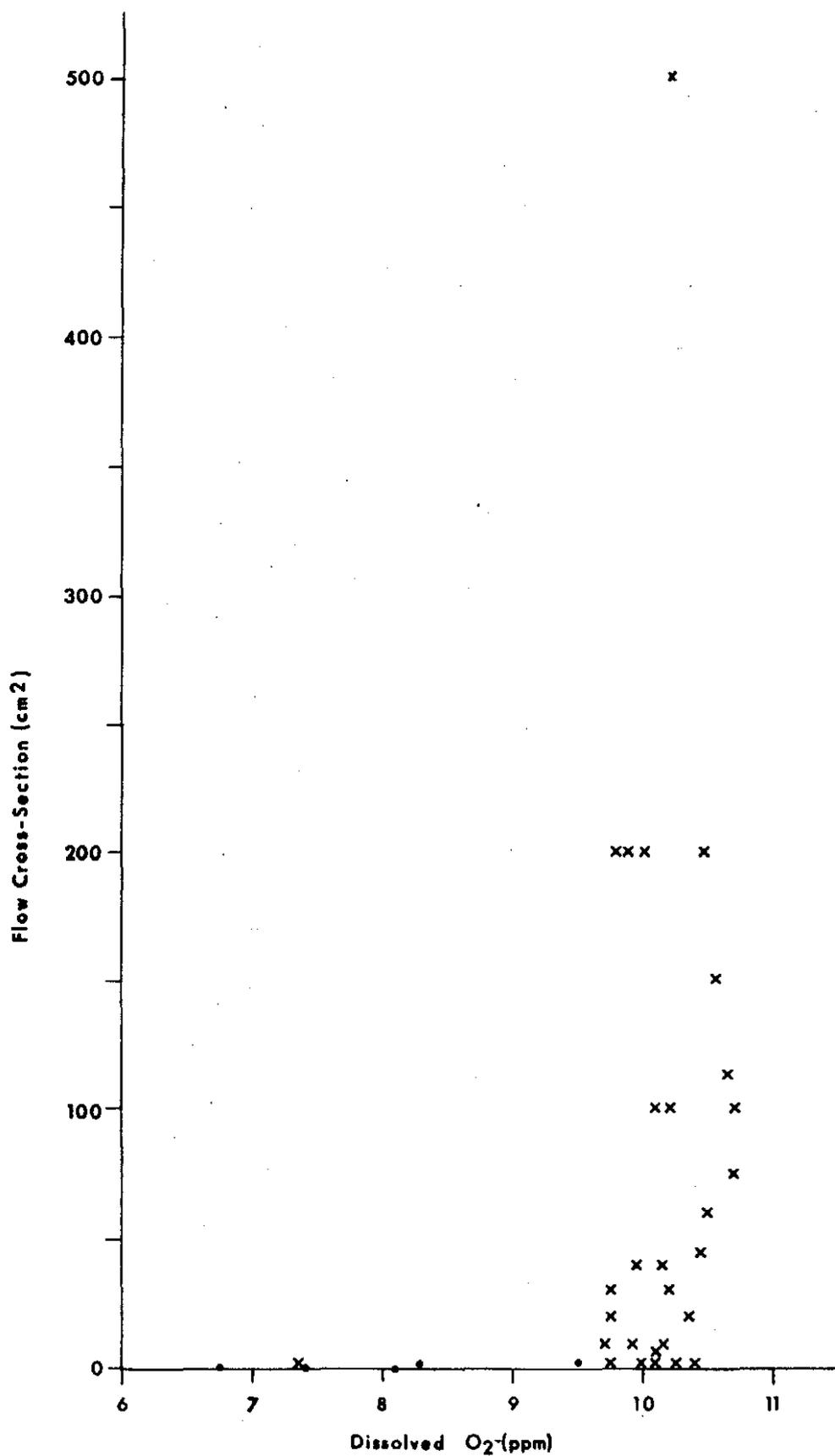


LOCATION MAP



THE BROKEN HILL PROPRIETARY CO. LTD. EXPLORATION DEPARTMENT		
MONTHLY AVERAGE RAINFALL NW. TASMANIA		
Prepared by: G. WATHUFF	Centre: SYDNEY	
Date: 25.11.80	Project No.	Drawing No.
Drawn: M. TSIANDARLIS		A3-

054



Flow Cross-Section vs Dissolved O₂

• pH < 6.0
x pH > 6.0

Centre:
SYDNEY

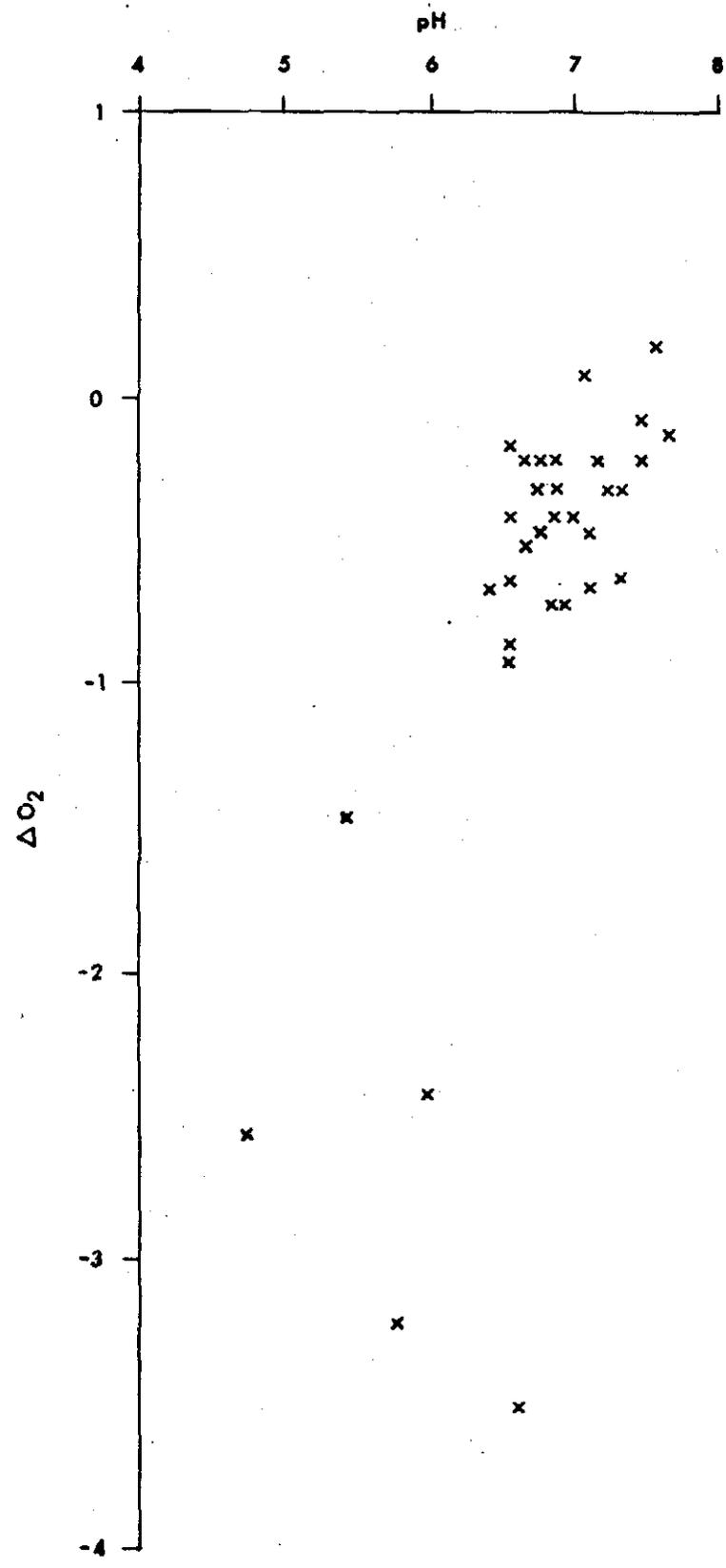
THE BROKEN HILL PROPRIETARY CO. LTD.

Project No:
T650, T640

Date:
19-04-1982

WARATAH-WYNYARD SPRING WATER

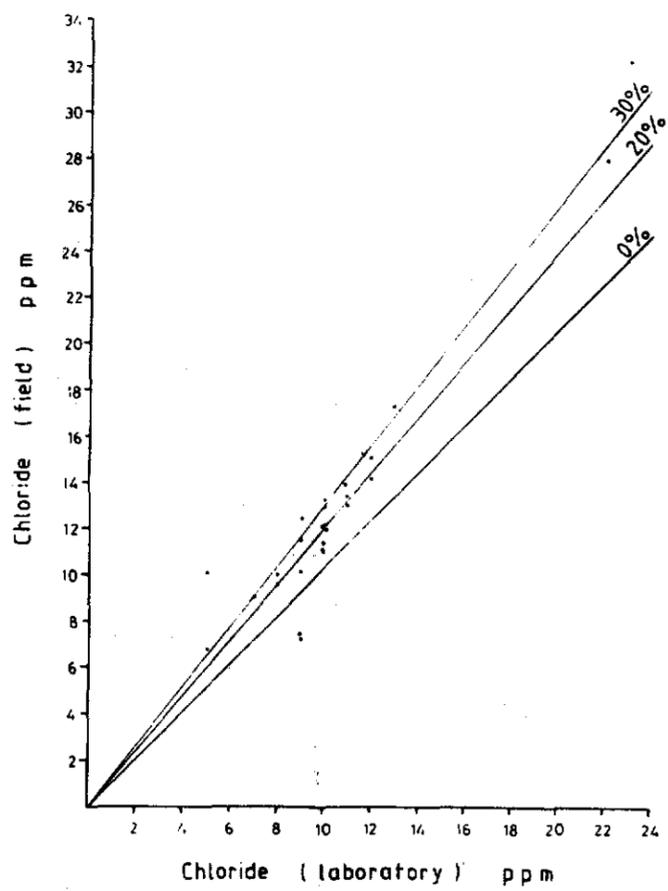
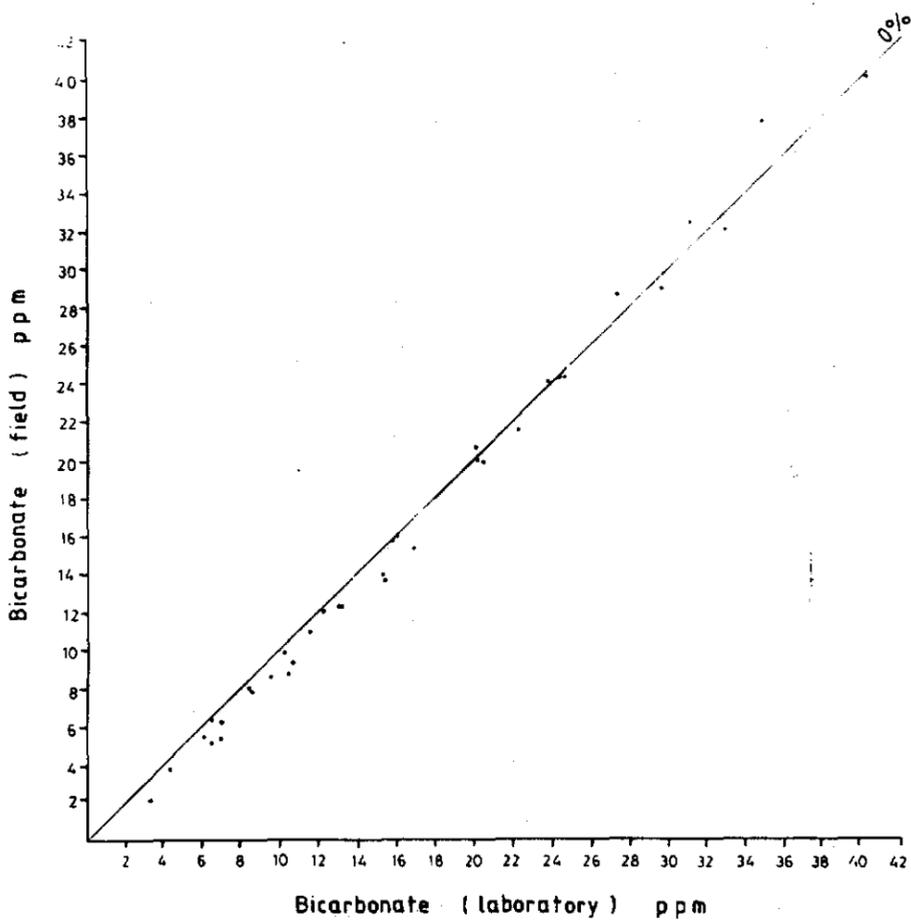
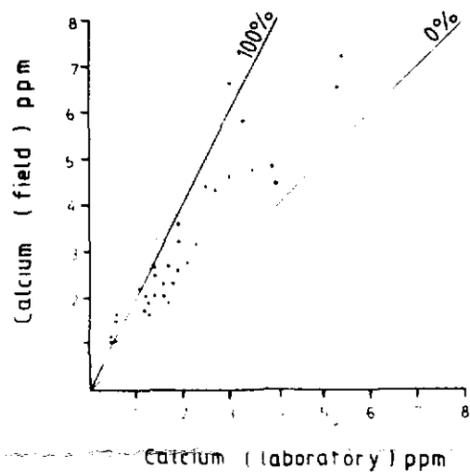
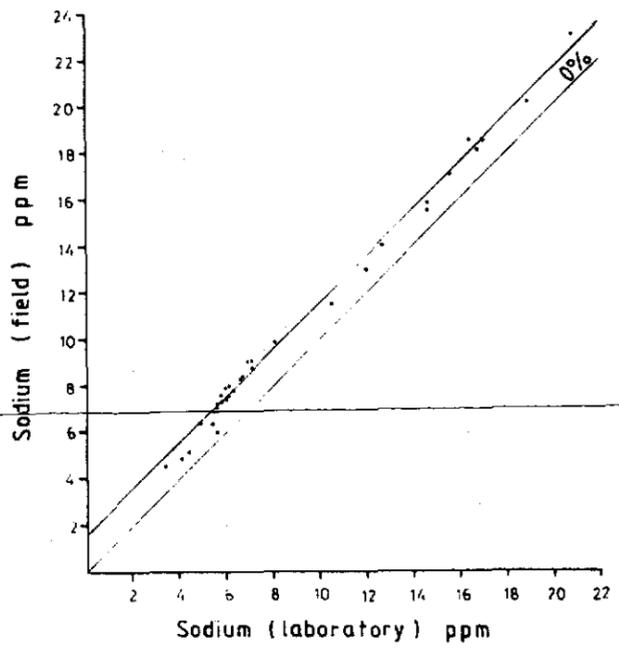
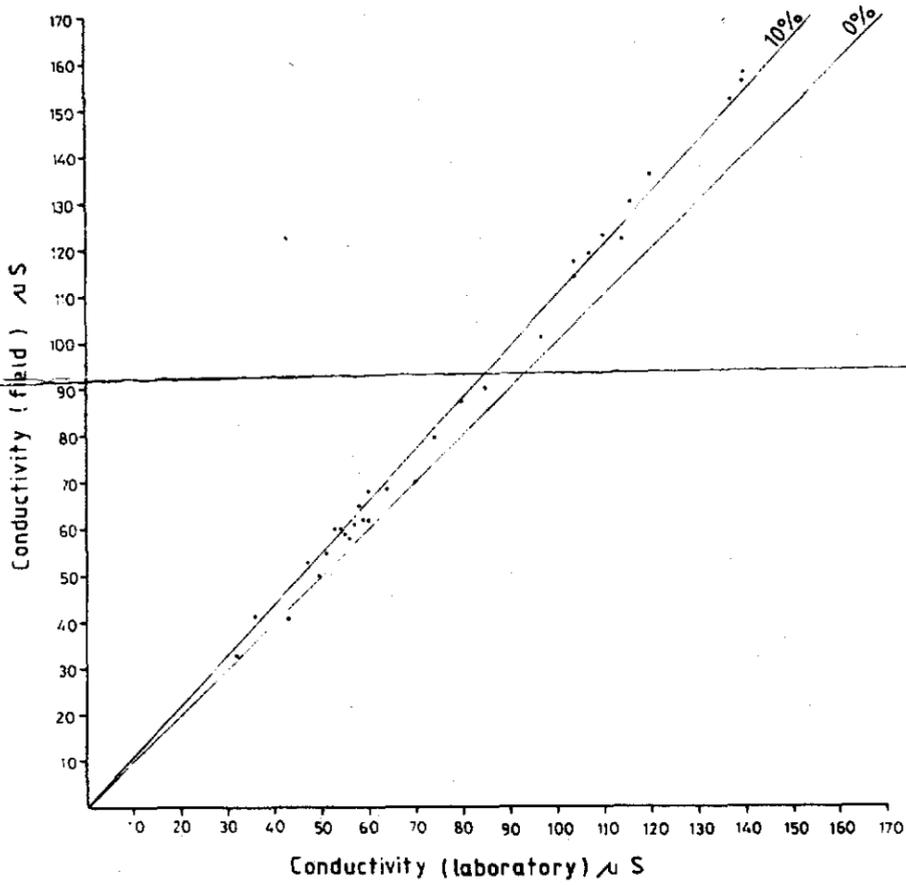
Drawing No:
A4-



ΔO₂ vs pH
 (ΔO₂ = O₂ measured (ppm) - O₂ at saturation (ppm))

055

277057 Fig. 4



THE BROKEN HILL PROPRIETARY CO. LTD. EXPLORATION DEPARTMENT		
Field vs Laboratory Determinations		
Conductivity Ca Na Hco ₃ Cl		
Drawn: G W	Date: 10 June 82	Centre: SYDNEY
Traced: O S	Project NO	Drawing NO
Checked:		A2-

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277058

FIG. 5

LEGEND

Dominant rock type at sample site

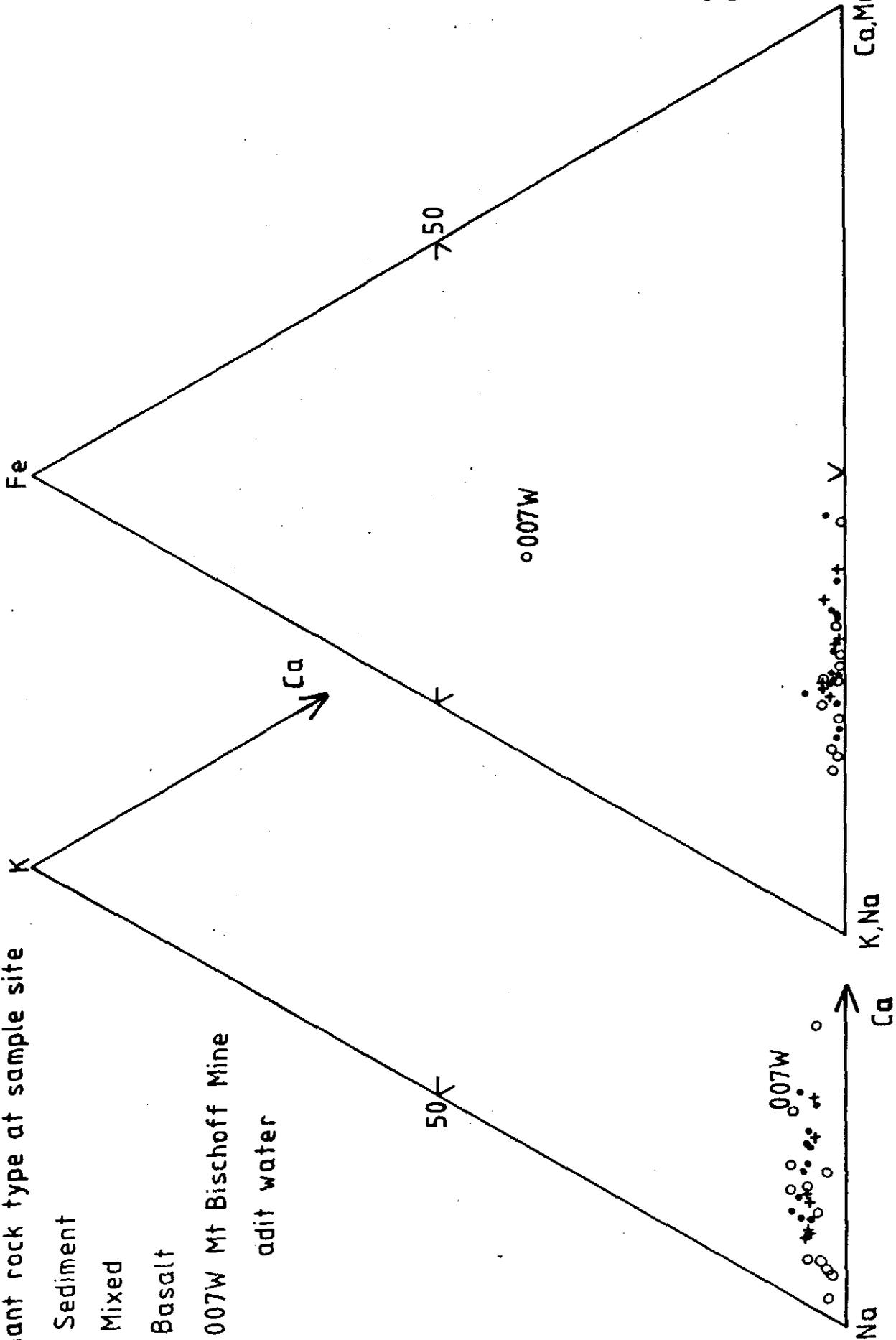
o Sediment

• Mixed

+ Basalt

o 007W Mt Bischoff Mine

adit water



Centre
 SYD
 Date
 12.1.6/82

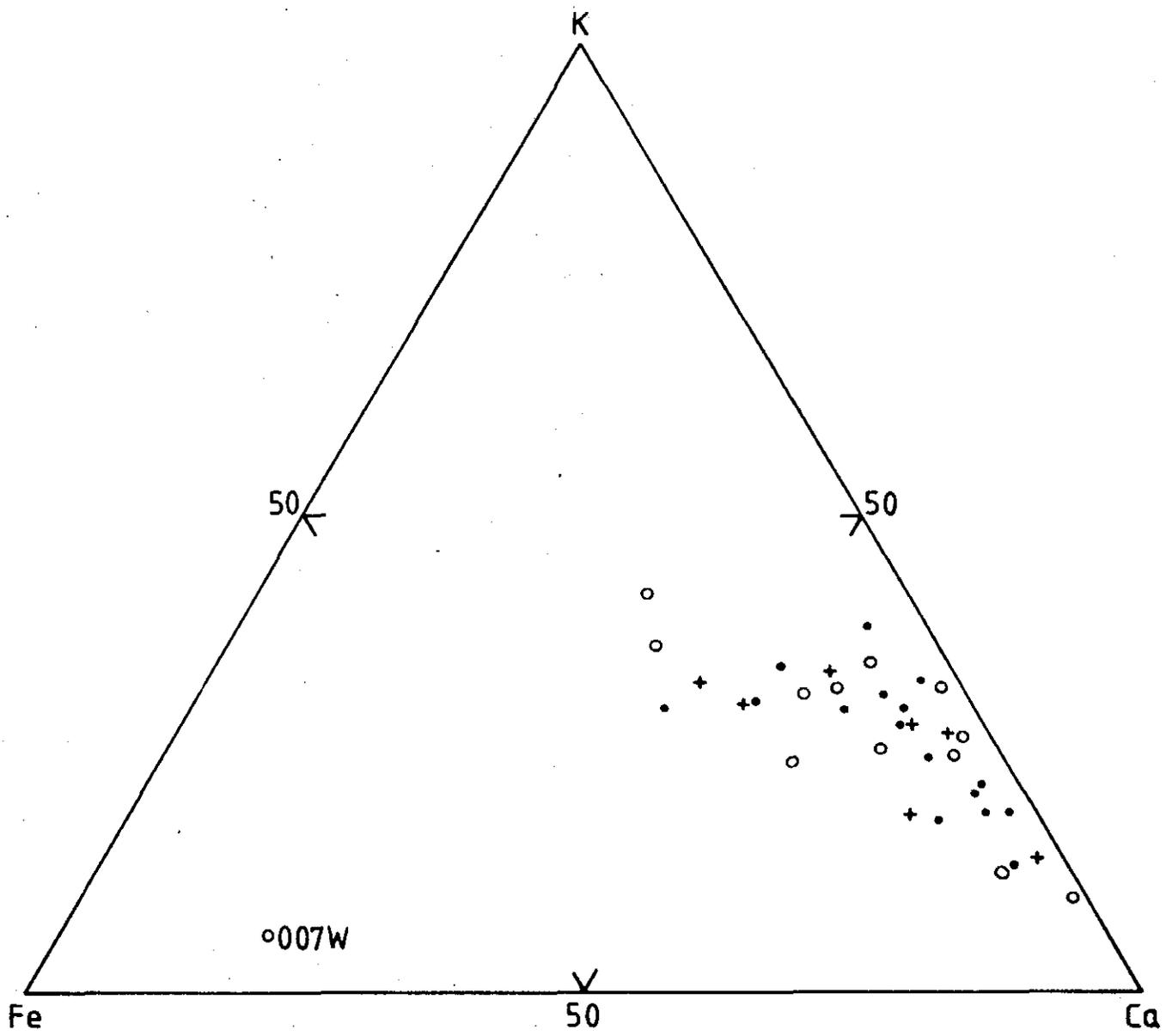
THE BROKEN HILL PROPRIETARY CO. LTD.
 EL 23 and 33/79 WYNYARD and WARATAH
 ATOMIC Na-K-Ca and AFM PLOTS FOR WATERS

Project No.
 Drawing No.

058

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FIG. 6

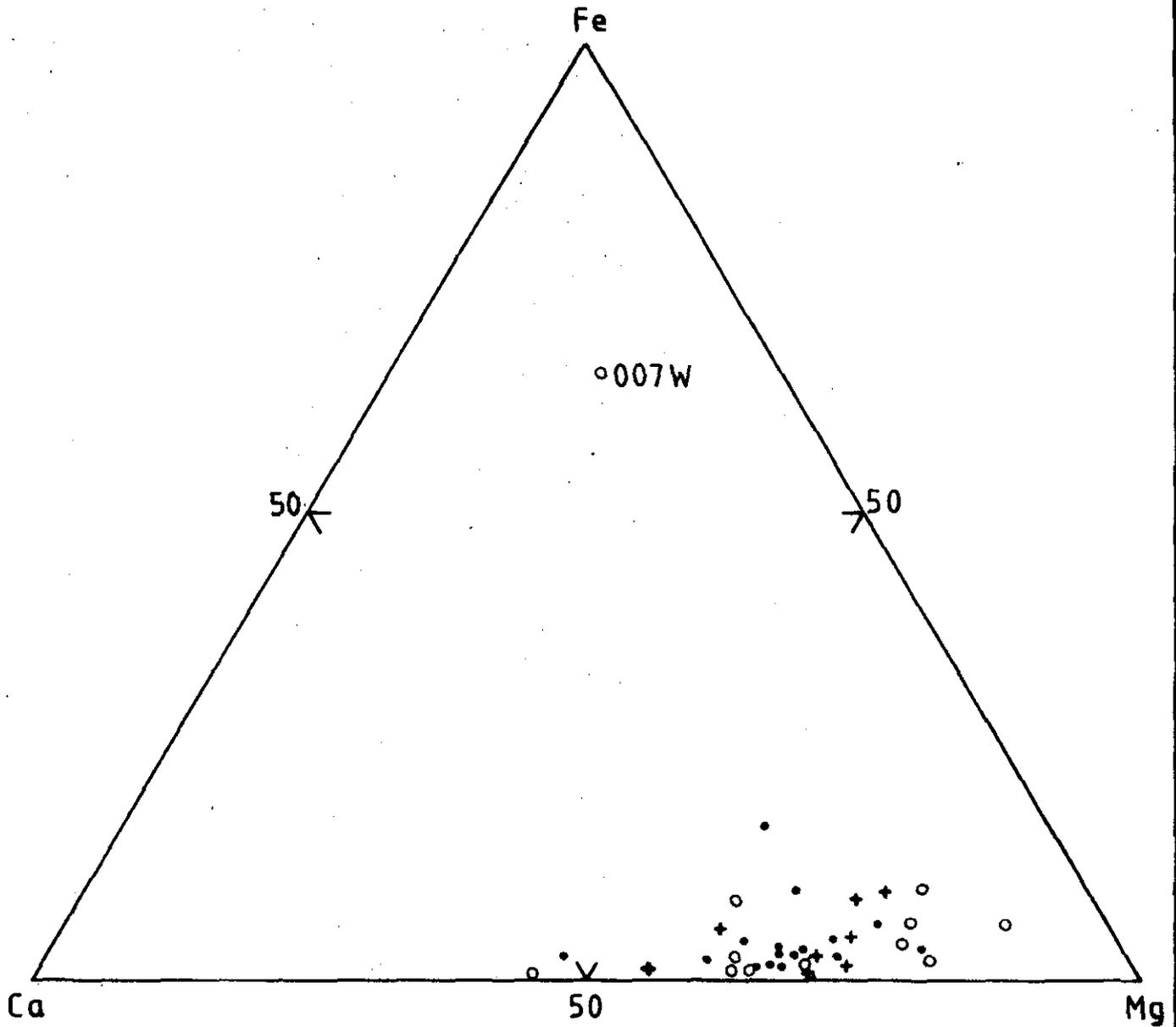


LEGEND

Dominant rock type at sample site

- Sediment
- Mixed
- + Basalt
- 007W Mt Bischoff Mine adit water

059



LEGEND

Dominant rock type at sample site

- Sediment
- Mixed
- + Basalt
- 007W Mt Bischoff Mine adit water

Centre SYD
 Date 12/6/82

THE BROKEN HILL PROPRIETARY CO. LTD.

EL 23/79 and 33/79 WYNYARD and WARATAH
 ATOMIC Ca-Fe-Mg PLOT FOR WATER SAMPLES

Project No.
 Drawing No.

APPENDIX 1

SAMPLE DESCRIPTIONS.

Sample No	Site	Temp.	pH	Cond.	O ₂	Free CO ₂	Ca	Cl	Na	Alk	HCO ₃ '	CO ₃ "	CO ₂ tot	Comments
026W	Field	10	6.5	122	10.15		2.02	27.9	17.0	5.6	10.6	0		20 x 2 cm flow 6/4/8 Tree ferns, sarrasiferae
	Lab.		6.5	114			1.4	22	15.6		9.3			Sediment + some basalt fragments Trace vein quartz.
027W	Field	11	5.4	130	9.5		1.04		18.5	1.1	2.1	0		2 x 1 cm flow 6/4/8 Tree ferns, sarrasiferae
	Lab.		5.8 _g	116			0.6	21	16.4		3.3			Sediment fragments + outcrop
028W	Field	12	7.4	119	10.5		2.48	17.3	15.5	24.4	29.0	0		20 x 3 cm flow. Sample taken 6/4 just below water fall
	Lab.		7.1 _o	107			1.4	13	14.6		29.6			Sediment outcrop Tree ferns, sarrasiferae, eucalypts
029W	Field	12	6.5	156	9.75		2.56		20.1	17.1	20.5	0		30 x 1 cm flow 6/4 Sediment, quartz fragments
	Lab.		6.5 _g	140			1.9	21	18.9		20.0			Tree ferns, eucalypts
030W	Field	12	7.05	158	9.95		2.32		23.0	20.6	24.2	0		20 x 2 cm flow (approx) 6/4 Sediment fragments
	Lab.		7.0 ₃	140			1.8	22	20.8		24.4			Tree ferns, sarrasiferae
031W	Field	11	7.2	123	10.55		4.85		13.0	23.7	28.7	0		100 x 1.5 cm flow 7/4/8 Basalt, sediment fragments
	Lab.		7.2 _o	110			3.9	13	12.0		27.3			Sediment outcrop Tree ferns
032W	Field	11	6.6	170	7.35		6.53		18.1	19.7	24.0	0		3 x 1 cm flow 7/4/8 Sediment outcrop
	Lab.		7.0 _o	154			5.3	23	16.7		23.8			Tree ferns
033W	Field	11	4.7	136	8.3		1.74	32.2	15.8	0	0	0		3 x 1 cm flow 7/4/8 Quartz, sediment (E quartz) fragment
	Lab.		4.2 _g	120			1.2	23	14.6	Acidity 10.8				Tree ferns, eucalypts, sarrasiferae
034W	Field	11	6.8	117.5	10.15		3.16		14.0	12.7	15.3	0		5 x 2 cm flow 7/4/8 Sediment + quartz fragments
	Lab.		6.6 _g	104			2.3	16	12.7		16.8			Phyllite + outcrop
035W	Field	12.3	7.5	114	10.7		4.47		11.5	31.1	37.8	0		50 x 2 cm flow 7/4/8 Sediment, basalt, vein quartz
	Lab.		7.2 _o	104			4.0	12	10.5		34.8			fragments. Blackberry, tree fern eucalypts, bracken fern

°C micro-siemens ppm ppm ppm ppm ppm ppm ppm ppm

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APPENDIX 2.

LABORATORY ANALYSIS SHEETS.

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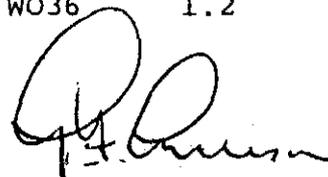
Our ref 29008/19

Dr G Watmuff
BHP Minerals Ltd
20 O'Connell Street
SYDNEY .. NSW .. 2000

ANALYSIS OF WATER SAMPLES

12 water samples contained in 1L (nominal) poly bottles and labelled as shown hereunder were received on 8/4/82. The contents were analysed by methods generally in accord with the APHA "Standard Methods for the Examination of Water and Wastewater" 14th Ed. (1975), details of which are available on request. The results of these analyses were as follows: (except for pH and conductivity, all figures are mgm/L)

	<u>K</u>	<u>Na</u>	<u>Cl⁻</u>	<u>pH</u>	<u>Cond</u> μ S	<u>HCO₃⁻</u>
WO26	0.7	15.6	22	6.5 ₂	114	7.6
WO27	0.7	16.4	21	5.8 ₈	116	2.7
WO28	1.3	14.6	13	7.1 ₀	107	24.3
WO29	1.1	18.9	21	6.5 ₈	140	16.4
WO30	1.0	20.8	22	7.0 ₃	140	20.0
WO31	1.1	12.0	13	7.2 ₀	110	22.4
WO32	0.8	16.7	23	7.0 ₀	154	19.5
WO33	0.5	14.6	23	4.2 ₅	120	NIL ACIDITY 10.8
WO34	0.9	12.7	16	6.6 ₉	104	13.8
WO35	1.1	10.5	12	7.2 ₆	104	28.5
WO36	1.2	17.0	19	7.2 ₁	137	25.5


G F ALLISON
CHARTERED CHEMIST (AUSTRALIA)



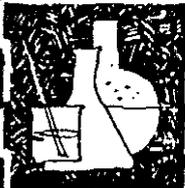
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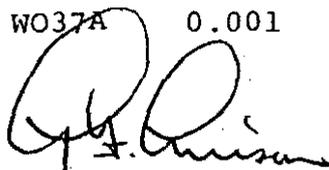
Our ref 29008/19

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BHP Minerals Ltd
20 O'Connell Street
SYDNEY .. NSW .. 2000

ANALYSIS OF WATER SAMPLES

12 water samples contained in 1L (nominal) poly bottles and labelled as shown hereunder were received on 8/4/82. The contents were analysed by methods generally in accord with the APHA "Standard Methods for the Examination of Water and Wastewater" 14th Ed. (1975), details of which are available on request. The results of these analyses were as follows: (except for pH and conductivity, all figures are mgm/L)

	<u>Cu</u>	<u>Pb</u>	<u>Zn</u>	<u>Fe</u>	<u>Ca</u>	<u>Mg</u>
WO26A	0.001	∕0.01	0.005	0.39	1.4	3.7
WO27A	∕0.001	∕0.01	0.006	0.56	0.6	3.4
WO28A	0.001	∕0.01	0.004	1.3	1.4	4.3
WO29A	∕0.001	∕0.01	0.084	0.32	1.9	5.1
WO30A	∕0.001	∕0.01	0.004	0.50	1.8	4.2
WO31A	0.001	∕0.01	0.003	0.25	3.9	4.5
WO32A	0.001	∕0.01	0.004	0.59	5.3	5.7
WO33A	0.001	∕0.01	0.005	0.58	1.2	3.1
WO34A	0.001	∕0.01	0.002	0.55	2.3	3.8
WO35A	∕0.001	∕0.01	0.003	0.33	4.0	4.9
WO36A	0.001	∕0.01	0.006	0.32	3.5	4.9
WO37A	0.001	∕0.01	0.005	0.008	∕0.1	∕0.1



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CHARTERED CHEMIST (AUSTRALIA)



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Our refs: 28174/84
28809/22
29008/19

Dr G Watmuff
Senior Geochemist
BHP Minerals Ltd
20 O'Connell Street
SYDNEY

ANALYSIS OF WATER SAMPLES - FLUORIDE

36 water samples contained in 1L poly bottles and labelled as shown hereunder were received 24 March, 2 April and 8 April 1982. The samples were analysed for fluoride by APHA "Standard Methods for the Examination of Water & Wastewater" 14th Ed. (1975), Method 414 C modified in this laboratory.

Results are expressed in mgm/L:

<u>LABEL</u>	<u>F⁻</u>
WO26	0.14
027	0.12
028	0.25
029	0.20
030	0.20
031	0.12
032	0.11
033	0.11
034	0.17
035	0.13
036	0.14

G F ALLISON

CHARTERED CHEMIST (AUSTRALIA)

REPORT NO 8981

ISSUED 31 August 1982



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- TANK AND CARGO SURVEYS
- EFFLUENT ANALYSIS

Our refs: 28174/84
28809/22
29008/19

Dr G Watmuff
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SYDNEY

ANALYSIS OF WATER SAMPLES - ARSENIC

From the thirtysix water samples labelled 001W to 036W received 24 March, 2 April and 8 April 1982, six were selected which were thought might have detectable arsenic levels. They were analysed in accord with APHA "Standard Methods for the Examination of Water & Wastewater" 14th Ed. (1975), Method 404A with the following results:

Results are expressed in mgm/L

<u>LABEL</u>	<u>ARSENIC</u>
W028	less than 0.02
W032	less than 0.02

G F ALLISON
CHARTERED CHEMIST (AUSTRALIA)

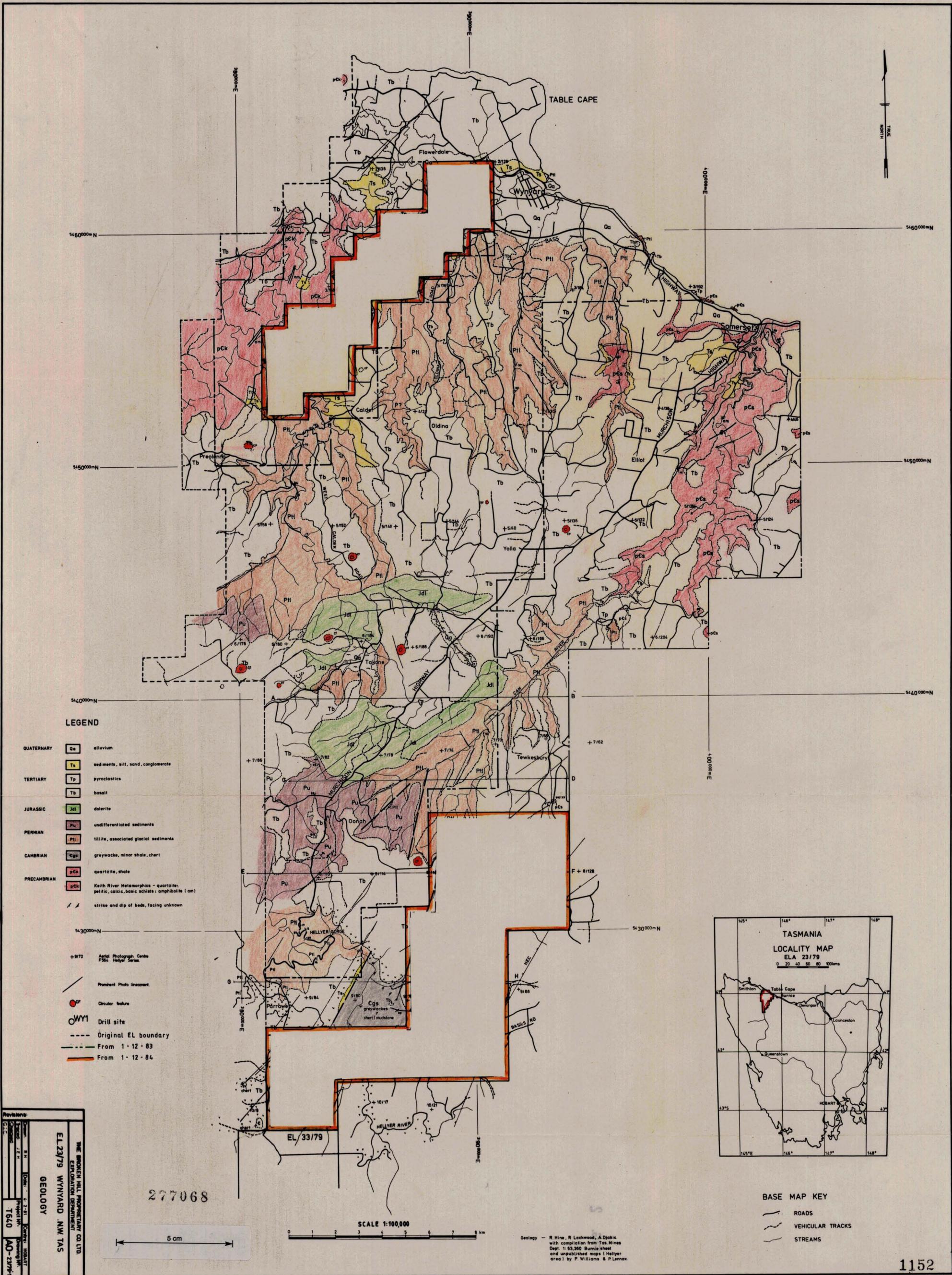
REPORT NO 8982

ISSUED 31 August 1982



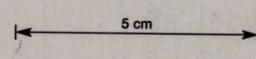
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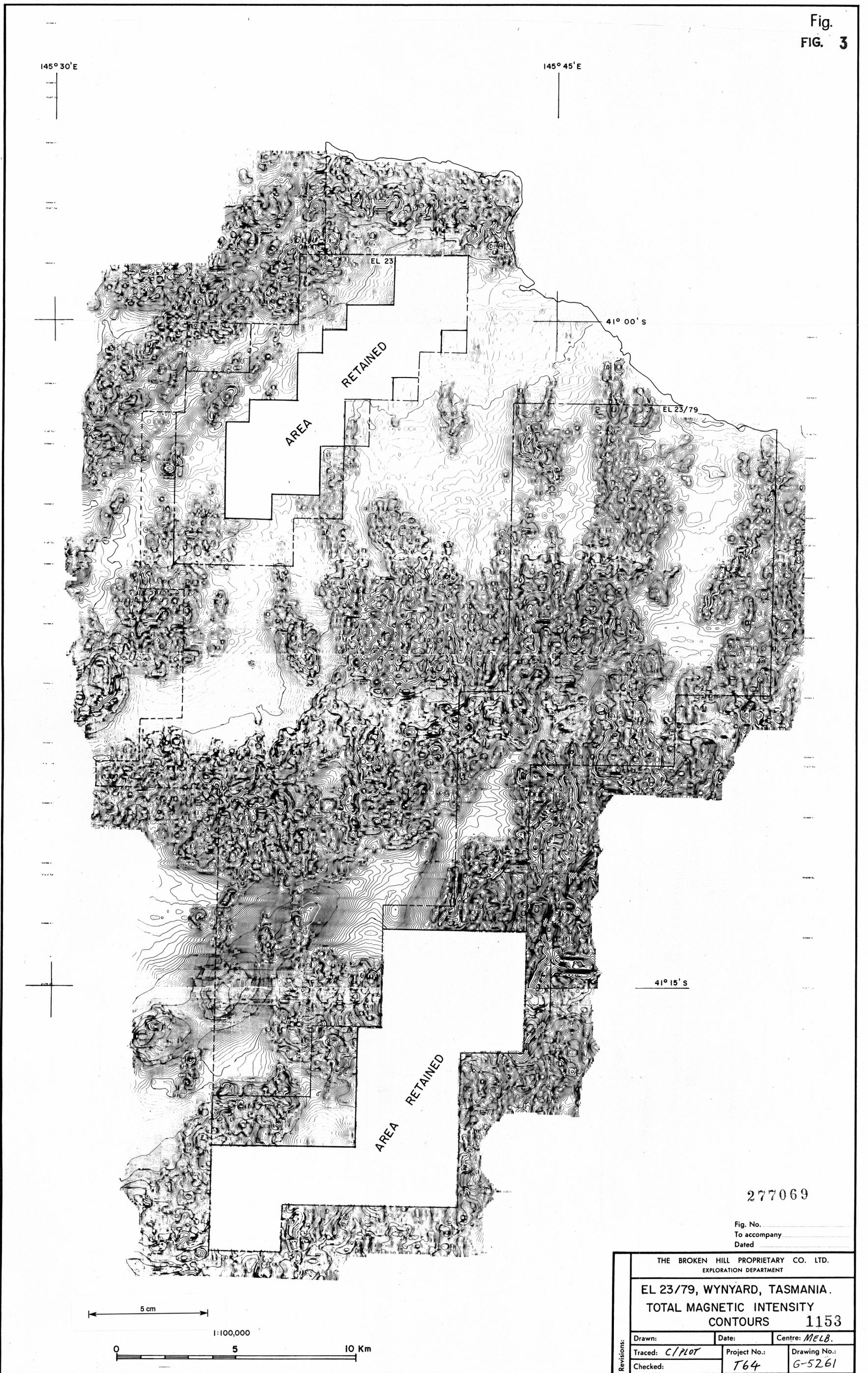
277068

SCALE 1:100,000



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 EXPLORATION DEPARTMENT
GEOLOGY
 EL 23/79 WYNYARD, NW TAS
 Project No. T640
 AD-2379-1

Fig. 3
FIG. 3



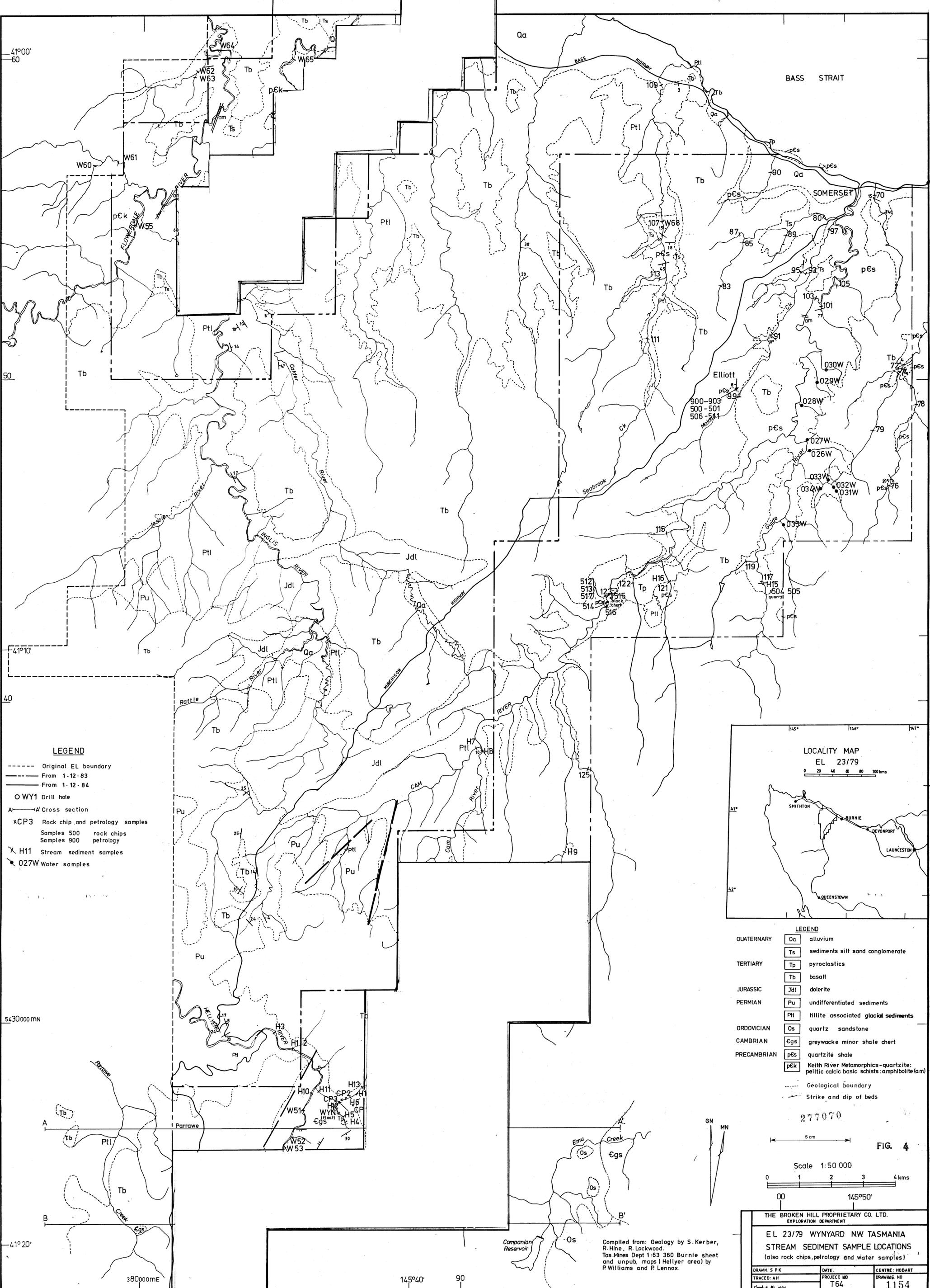
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Fig. No.
To accompany
Dated

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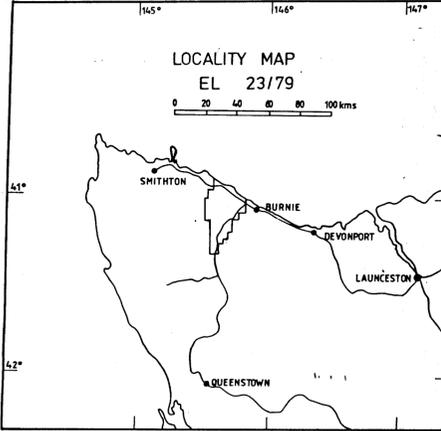
EL 23/79, WYNYARD, TASMANIA.
TOTAL MAGNETIC INTENSITY
CONTOURS 1153

Revisions:	Drawn:	Date:	Centre: MELB.
	Traced: C/PLOT	Project No.:	Drawing No.:
	Checked:	T64	G-5261



LEGEND

- Original EL boundary
- From 1-12-83
- From 1-12-84
- WY1 Drill hole
- A—A' Cross section
- xCP3 Rock chip and petrology samples
 - Samples 500 rock chips
 - Samples 900 petrology
- X H11 Stream sediment samples
- 027W Water samples



- LEGEND**
- QUATERNARY
 - Qa alluvium
 - Ts sediments silt sand conglomerate
 - TERTIARY
 - Tp pyroclastics
 - Tb basalt
 - JURASSIC
 - Jdl dolerite
 - PERMIAN
 - Pu undifferentiated sediments
 - ORDOVICIAN
 - Ptl tillite associated glacial sediments
 - ORDOVICIAN
 - Os quartz sandstone
 - CAMBRIAN
 - Cgs greywacke minor shale chert
 - PRECAMBRIAN
 - pCs quartzite shale
 - pEk Keith River Metamorphics-quartzite; pelitic calcic basic schists; amphibolite (am)
- Geological boundary
 --- Strike and dip of beds

277070

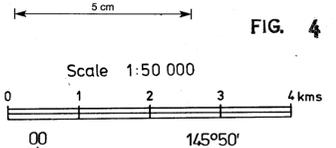
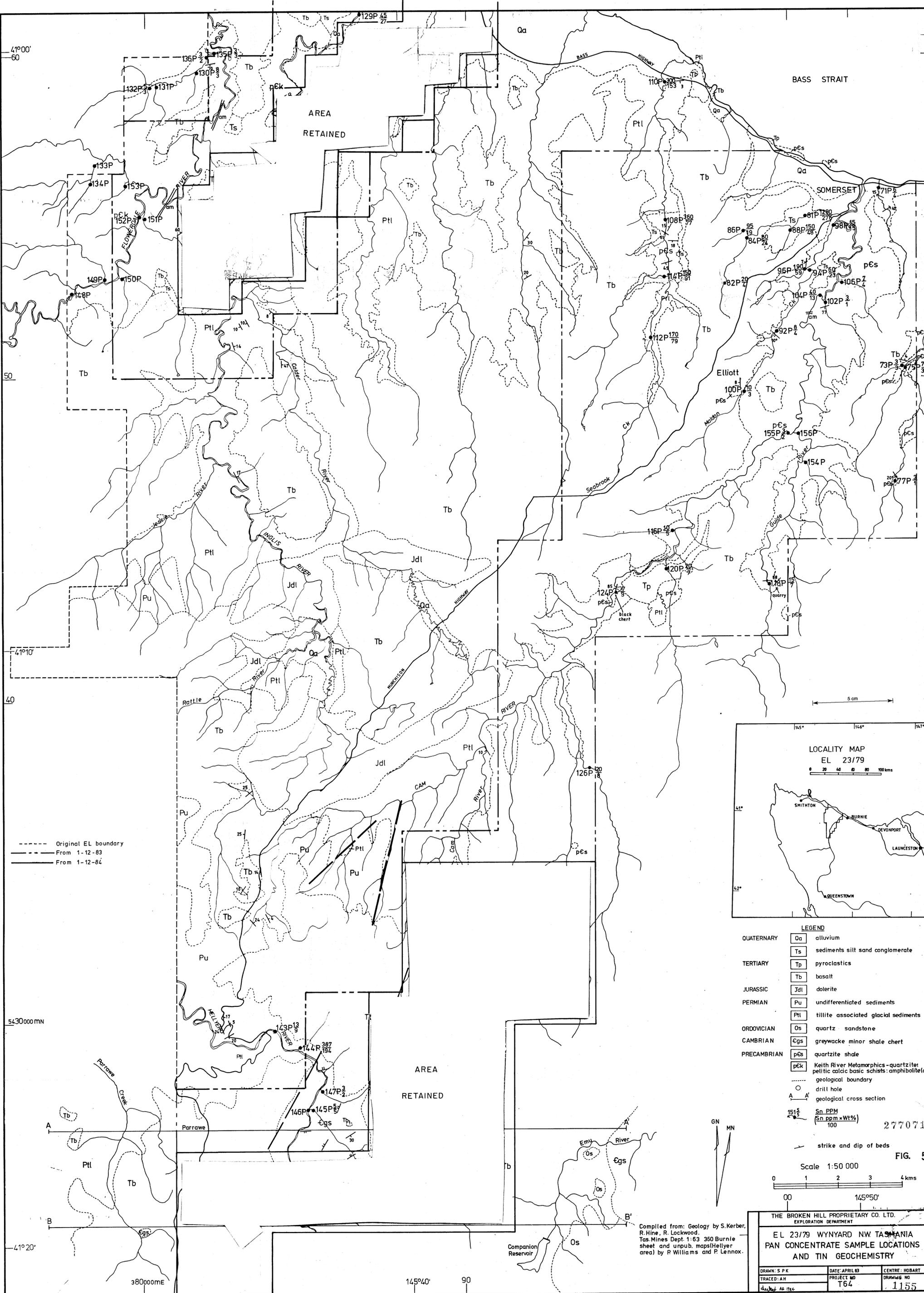


FIG. 4

THE BROKEN HILL PROPRIETARY CO. LTD. EXPLORATION DEPARTMENT		
EL 23/79 WYNARD NW TASMANIA STREAM SEDIMENT SAMPLE LOCATIONS (also rock chips, petrology and water samples)		
DRAWN: S P K	DATE:	CENTRE: HOBART
TRACED: AH	PROJECT NO: T64	DRAWING NO: 1154

Compiled from: Geology by S. Kerber, R. Hine, R. Lockwood. Tas. Mines Dept 1:63 360 Burnie sheet and unpub. maps (Hellyer area) by P. Williams and P. Lennox.



BASS STRAIT

SOMERSET

LOCALITY MAP
EL 23/79

0 20 40 60 80 100 kms

SMITHTON BURNIE DEVONPORT LAUNCESTON QUEENSTOWN

LEGEND

QUATERNARY	Qa	alluvium
	Ts	sediments silt sand conglomerate
TERTIARY	Tp	pyroclastics
	Tb	basalt
JURASSIC	Jdl	dolerite
PERMIAN	Pu	undifferentiated sediments
	Ptl	tillite associated glacial sediments
ORDOVICIAN	Os	quartz sandstone
CAMBRIAN	Cgs	greywacke minor shale chert
PRECAMBRIAN	pCs	quartzite shale
	pCk	Keith River Metamorphics-quartzite pelitic calcic basic schists amphibolite(am)
	○	drill hole
	—	geological boundary
	—	geological cross section
	—	strike and dip of beds

Scale 1:50 000

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EXPLORATION DEPARTMENT

EL 23/79 WYNARD NW TASMANIA
PAN CONCENTRATE SAMPLE LOCATIONS
AND TIN GEOCHEMISTRY

DRAWN: S.P.K.	DATE: APRIL 83	CENTRE: HOBBART
TRACED: A.H.	PROJECT NO: T64	DRAWING NO: 1155

Compiled from: Geology by S. Kerber, R. Hine, R. Lockwood, Tas. Mines Dept. 1:63 360 Burnie sheet and unpub. maps (Hellyer area) by P. Williams and P. Lennox.

FIG. 5

277071

--- Original EL boundary
- - - From 1-12-83
— From 1-12-84

5430000 MN

41° 20'

145° 40' 90

3800000 ME